

Naming Acids and Bases Answers

Name the following acids and bases:

- | | | |
|----|--------------------------------|-----------------------------|
| 1) | NaOH | <u>sodium hydroxide</u> |
| 2) | H ₂ SO ₃ | <u>sulfurous acid</u> |
| 3) | H ₂ S | <u>hydrosulfuric acid</u> |
| 4) | H ₃ PO ₄ | <u>phosphoric acid</u> |
| 5) | NH ₃ | <u>ammonia</u> |
| 6) | HCN | <u>hydrocyanic acid</u> |
| 7) | Ca(OH) ₂ | <u>calcium hydroxide</u> |
| 8) | Fe(OH) ₃ | <u>iron (III) hydroxide</u> |
| 9) | H ₃ P | <u>hydrophosphoric acid</u> |

Write the formulas of the following acids and bases:

- | | | |
|-----|-----------------------|------------------------------------|
| 10) | hydrofluoric acid | <u>HF</u> |
| 11) | hydroselenic acid | <u>H₂Se</u> |
| 12) | carbonic acid | <u>H₂CO₃</u> |
| 13) | lithium hydroxide | <u>LiOH</u> |
| 14) | nitrous acid | <u>HNO₂</u> |
| 15) | cobalt (II) hydroxide | <u>Co(OH)₂</u> |
| 16) | sulfuric acid | <u>H₂SO₄</u> |
| 17) | beryllium hydroxide | <u>Be(OH)₂</u> |
| 18) | hydrobromic acid | <u>HBr</u> |

Answers For Naming Acids

RM Cervero



Answers For Naming Acids:

Homework Helpers: Chemistry Greg Curran, 2025-09-12 Homework Helpers Chemistry is a user friendly review book that will make every student or parent trying to help their child feel like he or she has a private Chemistry tutor Concepts are explained in clear easy to understand language and problems are worked out with step by step methods that are easy to follow Each lesson comes with numerous review questions and answer keynotes that explain each correct answer and why it is correct This book covers all of the topics in a typical one year Chemistry curriculum including A systematic approach to problem solving conversions and the use of units Naming compounds writing formulas and balancing chemical equations Gas laws chemical kinetics acids and bases electrochemistry and more While Homework Helpers Chemistry is an excellent review for any standardized Chemistry test including the SAT II its real value is in providing support and guidance during the year's entire course of study

Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced eText with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

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discussed in the chapter **Chemistry: Concepts and Problems** Clifford C. Houk, Richard Post, 1996-03-09 CHEMISTRY SECOND EDITION The fast easy way to master the fundamentals of chemistry Have you ever wondered about the differences between liquids gases and solids Or what actually happens when something burns What exactly is a solution An acid A base This is chemistry the composition and structure of substances composing all matter and how they can be transformed Whether you are studying chemistry for the first time on your own want to refresh your memory for a test or need a little help for a course this concise interactive guide gives you a fresh approach to this fascinating subject This fully up to date edition of Chemistry Concepts and Problems Has been tested rewritten and retested to ensure that you can teach yourself all about chemistry Requires no prerequisites Lets you work at your own pace with a helpful question and answer format Lists objectives for each chapter you can skip ahead or find extra help if you need it Reinforces what you learn with chapter self tests

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Introduction to Chemistry Amos Turk, 2013-07-15 Introduction to Chemistry is a 26 chapter introductory textbook in general chemistry This book deals first with the atoms and the arithmetic and energetics of their combination into molecules The subsequent chapters consider the nature of the interactions among atoms or the so called chemical bonding This topic is followed by discussions on the nature of intermolecular forces and the states of matter This text further explores the statistics and dynamics of chemistry including the study of equilibrium and kinetics Other chapters cover the aspects of ionic equilibrium acids and bases and galvanic cells The concluding chapters focus on a descriptive study of chemistry such as the representative and transition elements organic and nuclear chemistry metals polymers and biochemistry Teachers and undergraduate chemistry students will find this book of great value **A Visual Analogy Guide to Chemistry, 2e** Paul A

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The Chemistry Student's Companion Stephen R. Schaffter,2006 Designed for an Honors Chemistry class this book covers all of the California State Standards for Chemistry Cover

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