Oxidation-Reduction Analysis of Hydrogen Peroxide

Many oxidation-reduction reactions serve as a basis for quantitative analysis. One of the most widely used oxidizing agents is potassium permanganate, KMnO4. It is a powerful oxidizing agent that also serves as a self-indicator. One drop in excess of the endpoint volume of a titration turns the solution a pale-pink color.

In this experiment, you will titrate a standardized solution of 0.02 M KMnO₄ into a measured amount of an unknown concentration of hydrogen peroxide, H₂O₂. Your objective is to determine the percentage and molarity of H₂O₂ in ordinary over-the-counter solutions of hydrogen peroxide.

Pre-Lab Assignment:

Balance the equation for the titration reaction using the half-reaction method. The reaction takes place in acidic conditions. (Record your work in your lab notebook.)

$$H_2O_2$$
 (aq) + MnO_4^- (aq) $\rightarrow O_2$ (g) + Mn^{2+} (aq)

Materials

2 burets
Erlenmeyer flasks
Stir plate
Magnetic stir bar
Balance
Household hydrogen peroxide
0.02M potassium permanganate solution
3.0 M sulfuric acid
0.1M manganese(II) sulfate
Distilled water

PROCEDURE

- 1. Obtain and wear goggles.
- Rinse and fill a buret with standardized KMnO4 solution. Record the molarity of the solution.
- Determine the mass of a clean, dry Erlenmeyer flask to the correct number of significant figures, and record.
- From the buret on the front table, record the initial volume (again, to the correct number of significant figures). Add approximately 1.5 mL of ordinary household hydrogen peroxide. Record the final buret volume.
- 5. Find the mass of the flask with the peroxide and record.
- Add about 35 mL of distilled water, 5 mL of 3.0 M H₂SO₄, and 3 or 4 drops of 0.1 M MnSO₄, which acts as a catalyst.

Analysis Of Hydrogen Peroxide Pre Lab Answer

K Morrison

Analysis Of Hydrogen Peroxide Pre Lab Answer:

Subject Index to Unclassified ASTIA Documents Defense Documentation Center (U.S.), 1960 Methods in Biotechnology Seung-Beom Hong, M. Bazlur Rashid, Lory Z. Santiago-Vázguez, 2016-08-01 As rapid advances in biotechnology occur there is a need for a pedagogical tool to aid current students and laboratory professionals in biotechnological methods Methods in Biotechnology is an invaluable resource for those students and professionals Methods in Biotechnology engages the reader by implementing an active learning approach provided advanced study guestions as well as pre and post lab questions for each lab protocol These self directed study sections encourage the reader to not just perform experiments but to engage with the material on a higher level utilizing critical thinking and troubleshooting skills This text is broken into three sections based on level Methods in Biotechnology Advanced Methods in Biotechnology I and Advanced Methods in Biotechnology II Each section contains 14 22 lab exercises with instructor notes in appendices as well as an answer guide as a part of the book companion site This text will be an excellent resource for both students and laboratory professionals in the biotechnology field Laboratory Experiments for General Chemistry Harold R. Hunt, Toby F. General Chemistry Hunt, 1976-06-07 Block, George M. McKelvy, 1998 **Experimental Organic Chemistry** John C. Gilbert, Stephen F. Martin, 2006 This proven and well tested laboratory manual for organic chemistry students contains procedures for both miniscale also known as small scale and microscale users This lab manual gives students all the necessary background to enter the laboratory with the knowledge to perform the experiments with confidence For the microscale labs experiments were chosen to provide tangible quantities of material which can then be analyzed Chapters 1 2 introduce students to the equipment record keeping and safety of the laboratory Chapters 3 6 and 8 are designed to introduce students to laboratory techniques needed to perform all experiments In Chapters 7 and 9 through 20 students are required to use the techniques to synthesize compounds and analyze their properties In Chapter 21 students are introduced to multi step syntheses of organic compounds a practice well known in chemical industry In Chapter 23 students are asked to solve structures of unknown compounds The new chapter 24 introduces a meaningful experiment into the textbook that reflects the increasing emphasis on bioorganic chemistry in the sophomore level organic lecture course This experiment not only gives students the opportunity to accomplish a mechanistically interesting and synthetically important coupling of two a amino acids to produce a dipeptide but also provides valuable experience regarding the role of protecting groups in effecting synthetic transformations with multiple functionalized molecules Integrated Approach to Coordination Chemistry Rosemary A. Marusak, Kate Doan, Scott D. Cummings, 2007-03-07 Coordination chemistry is the study of compounds formed between metal ions and other neutral or negatively charged molecules This book offers a series of investigative inorganic laboratories approached through systematic coordination chemistry. It not only highlights the key fundamental components of the coordination chemistry field it also exemplifies the historical development of concepts in the field In order to graduate as

a chemistry major that fills the requirements of the American Chemical Society a student needs to take a laboratory course in inorganic chemistry Most professors who teach and inorganic chemistry laboratory prefer to emphasize coordination chemistry rather than attempting to cover all aspects of inorganic chemistry because it keeps the students focused on a cohesive part of inorganic chemistry which has applications in medicine the environment molecular biology organic synthesis and inorganic materials

Introductory Chemistry Steven S. Zumdahl, 1996

Nuclear Science Abstracts , 1970

Green Organic Chemistry Kenneth M. Doxsee, James E. Hutchison, 2004 This lab text describes the tools and strategies of green chemistry and the lab experiments that allow investigation of organic chemistry concepts and techniques in a greener laboratory setting Students acquire the tools to assess the health and environmental impacts of chemical processes and the strategies to improve develop new processes that are less harmful to human health and the environment The curriculum introduces a number of state of the art experiments and reduces reliance on expensive environmental controls such as fume hoods Provided by publisher Characterization and Metrology for ULSI Technology 2005 David G. Seiler, 2005-09-29 The worldwide semiconductor community faces increasingly difficult challenges in the era of silicon nanotechnology and beyond The magnitude of these challenges demands special attention from the metrology and analytical measurements community New paradigms must be found Adequate research and development for new metrology concepts are urgently needed Characterization and metrology are key enablers for developing new semiconductor technology and in improving manufacturing This book summarizes major issues and gives critical reviews of important measurement techniques that are crucial to continuing the advances in semiconductor technology. It covers major aspects of process technology and most characterization techniques for silicon research including development manufacturing and diagnostics The book also covers emerging nano devices and the corresponding metrology challenges that arise **Journal of Solar Energy Engineering** ,1996 Watts' Dictionary of Chemistry Henry Watts, 1894 Energy Research Abstracts ,1994 **Environment Information Access** ,1972 American Industrial Hygiene Association Journal American Industrial Hygiene Association, 1961 Government Reports Announcements & Index ,1977 Chemical Abstracts ,1907 Pollution Abstracts, 1986 Indexes material from conference proceedings and hard to find documents in addition to journal articles Over 1 000 journals are indexed and literature published from 1981 to the present is covered Topics in pollution and its management are extensively covered from the standpoints of atmosphere emissions mathematical models effects on people and animals and environmental action Major areas of coverage include air pollution marine pollution freshwater pollution sewage and wastewater treatment waste management land pollution toxicology and health noise and Journal of the Air Pollution Control Association ,1972 radiation **Archives of Pathology & Laboratory** Medicine .2004

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Draw Lewis structures for each of the following. 1. NO3. 1 2. CH4. Skill Practice 26 Skill Practice 26. Name: Date:
Hour: 1. What does it mean to say that a bond is polar? One of the atoms Skill Practice 16 - Atomic Size Skill
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