

MOLAR SPECIFIC HEAT OF GASES:

1. $\Delta U = C_V \Delta T$
2. $C_p - C_v = R$
3. $C_v = R/\gamma - 1$
4. $C_p = \gamma R/\gamma - 1$
5. $C_v = 1/2 f R$ where f = degree of freedom
6. $C_p = (f + 2/2)R$
7. $\gamma = C_p/C_v$

THERMODYNAMIC SCALE:

1. $T = 273.16 Q_3/Q_2$
2. $Q_2/Q_3 = T_2/273.16 K$
3. $\frac{Q_1 - Q_2}{Q_2} = \frac{P_1 - P_2}{P_2} = \frac{V_1 - V_2}{V_2}$
4. $\frac{\Delta T}{100} = \frac{\Delta T}{100} = \frac{\Delta T}{100}$

2nd year MDCAT FORMULA LIST PHYSICS

ELECTROSTATICS

COULOMB'S LAW:

1. $F = k \frac{Q_1 Q_2}{r^2}$
2. $k = \frac{1}{4\pi\epsilon_0}$
3. $F = \frac{1}{4\pi\epsilon_0} \frac{Q_1 Q_2}{r^2}$
4. $E = \frac{F}{q}$
5. $\epsilon_0 = \epsilon_r \epsilon_0$
6. $F = \frac{1}{4\pi\epsilon_0} \frac{Q_1 Q_2}{r^2}$
7. $F = \frac{Q_1 Q_2}{4\pi\epsilon_0 r^2}$

ELECTRIC FIELD INTENSITY:

1. $E = \frac{F}{q}$
2. $E = \frac{1}{4\pi\epsilon_0} \frac{Q}{r^2}$
3. $E = \frac{1}{4\pi\epsilon_0} \frac{Q}{r^2}$
4. $E = \frac{1}{4\pi\epsilon_0} \frac{Q}{r^2}$
5. $E = \frac{Q}{4\pi\epsilon_0 r^2}$
6. $E = \frac{Q}{4\pi\epsilon_0 r^2}$
7. $E = \frac{Q}{4\pi\epsilon_0 r^2} = E = \frac{Q}{4\pi\epsilon_0 r^2}$

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