

Alkanes, Alkenes, Alkynes and Cyclic Hydrocarbons - Worksheet - ANSWERS

Name or draw the following compounds:

	Chemical structure	IUPAC Name
1	$\text{H}_3\text{C}-\text{CH}_2-\overset{\text{Cl}}{\underset{ }{\text{CH}}}-\text{CH}_3$	2-chlorobutane
2	$\begin{array}{c} \text{CH}_3 \\ \\ \text{H}_3\text{C}-\text{CH}-\text{CH}_2-\text{HC}-\text{CH}_3 \\ \quad \\ \text{CH}_3 \text{CH}_3 \end{array}$	3,5-dimethylheptane
3	$\text{H}_3\text{C}-\overset{\text{Br}}{\underset{ }{\text{CH}}}-\text{CH}_2-\overset{\text{Br}}{\underset{ }{\text{CH}}}-\underset{\text{CH}_3}{\underset{ }{\text{CH}}}-\underset{\text{Cl}}{\underset{ }{\text{C}}}-\underset{\text{Cl}}{\underset{ }{\text{C}}}-\text{CH}_3$	4,6-dibromo-2,2-dichloro-3-methylheptane
4	$\begin{array}{c} \text{CH}_3 \\ \\ \text{H}_3\text{C}-\text{C}-\text{CH}_2-\underset{\text{NH}_2}{\underset{ }{\text{CH}}}-\text{CH}_2-\text{CH}_2-\text{NO}_2 \\ \\ \text{CH}_3 \end{array}$	3-amino-5-ethyl-5-methyl-1-nitroheptane
5	$\text{H}_3\text{C}-\underset{\text{CH}_3}{\underset{ }{\text{CH}}}-\text{CH}_2-\underset{\text{CH}_3}{\underset{ }{\text{C}}}-\underset{\text{CH}_3}{\underset{ }{\text{C}}}-\text{CH}_2-\text{CH}_3$	2,4,4-trimethylhexane
6	$\begin{array}{c} \text{CH}_3 \\ \\ \text{HC}-\text{CH}_2-\underset{\text{Cl}}{\underset{ }{\text{CH}}}-\underset{\text{Cl}}{\underset{ }{\text{CH}}}-\text{CH}_3 \\ \\ \text{CH}_2 \\ \\ \text{Cl}-\underset{\text{CH}_3}{\underset{ }{\text{CH}}} \end{array}$	2,3,7-trichloro-5-ethyloctane
7	$\text{H}_3\text{C}-\underset{\text{NO}_2}{\underset{ }{\text{C}}}(\text{NH}_2)-\text{CH}_2-\underset{\text{NO}_2}{\underset{ }{\text{C}}}(\text{NH}_2)-\text{CH}_3$	2,4-diamino-2,4-dinitropentane
8	$\text{Cl}-\underset{\text{Cl}}{\underset{ }{\text{C}}}(\text{Cl})-\text{CH}_2-\underset{\text{Cl}}{\underset{ }{\text{C}}}(\text{Cl})-\text{CH}_2-\underset{\text{Cl}}{\underset{ }{\text{C}}}(\text{Cl})-\text{CH}_3$	1,1,1,3,3,5,5-heptachlorohexane
9	$\text{H}_3\text{C}-\underset{\text{CH}_3}{\underset{ }{\text{C}}}(\text{CH}_3)-\text{CH}_2-\overset{\text{Br}}{\underset{ }{\text{CH}}}-\text{CH}_3$	2-bromo-4,4-dimethylhexane
10	$\text{H}_3\text{C}-\text{CH}_2-\underset{\text{Cl}}{\underset{ }{\text{C}}}(\text{Br})-\text{CH}_2-\text{CH}_3$	3-bromo-3-chloropentane
11	$\text{H}_3\text{C}-\underset{\text{NO}_2}{\underset{ }{\text{CH}}}-\text{CH}_2-\overset{\text{Br}}{\underset{ }{\text{CH}}}-\text{CH}_3$	1-bromo-3-nitrobutane
12	$\text{CH}_3-\overset{\text{Cl}}{\underset{ }{\text{CH}}}-\underset{\text{Cl}}{\underset{ }{\text{CH}}}-\text{CH}_2-\text{CH}_3$	2,3-dichloropentane

Alkenes And Alkynes Study Guide Answers

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that has great relevance for all scientists not just chemists For chemical engineers understanding the properties of organic molecules and how reactions occur is critically important to understanding the processes in an industrial plant For biologists and health professionals it is essential because nearly all of biochemistry springs from organic chemistry Additionally all scientists can benefit from improved critical thinking and problem solving skills that are developed from the study of organic chemistry Organic chemistry like any skill is best learned by doing It is difficult to learn by rote memorization and true understanding comes only from concentrated reading and working as many problems as possible In fact problem sets are the best way to ensure that concepts are not only well understood but can also be applied to real world problems in the work place Helps readers learn to categorize analyze and solve organic chemistry problems at all levels of difficulty Hundreds of fully worked practice problems all with solutions Key concept summaries for every chapter reinforces core content from the companion book

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Study Guide, Introductory Chemistry, a Foundation ; Introductory Chemistry ; Basic Chemistry Iris Stovall,Donald J. DeCoste,2000

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