

A Wellness Guide to a Speedy Recovery and Healthy Living

Helena A. Proala, RN and Helena Proala, B.Pharm.



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web we utilize the equation that relate amount remaining initial mass and number of half lives $n = \frac{t}{t_{1/2}} = \frac{24 \times 50.3}{125} = 9.6$ what is the half life of an isotope that is 80% remained after 16 days solution remaining 80% 100% therefore $n = \frac{t}{t_{1/2}} = \frac{16}{100} = 0.16$ now using the half life equation in example 1 we have

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