

Balancing redox equations

Example

Balance the equation showing the oxidation of Fe^{2+} ions to Fe^{3+} ions by dichromate ions ($\text{Cr}_2\text{O}_7^{2-}$) in an acidic medium



Step 1: Identify oxidising and reducing agents and write half reactions



$$2\text{Cr} = 12$$

$$\text{Cr} = +6$$



Iron loses one electron, i.e. it is *oxidised*. It donates one electron to dichromate, i.e. it is the *reducing agent*.

Half equation: Oxidation reaction: $\text{Fe}^{2+} \rightarrow \text{Fe}^{3+} + \text{e}^-$



Chromium gains three electrons, i.e. it is *reduced*. It gains its electrons from iron, i.e. it is the *oxidising agent*.

Half equation: Reduction reaction: $\text{Cr}_2^{6+} + 3\text{e}^- \rightarrow \text{Cr}^{3+}$

Analyzing Oxidation Reduction Reactions Answers

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, **Electrochemical Methods for the Synthesis and Analysis of Advanced Functional Layers and Coatings** Stephan Vladymirov Kozhukharov, 2024-11-07 The content of this book is devoted to the synthesis of advanced nano materials and layers by application of electrochemical methods Examples for their applications in fuel cells batteries photovoltaics magnetic materials environmentally friendly anticorrosion coatings and surface pretreatments especially anodization are provided Cu metallization of non conductive substrates is proposed as well In addition electrochemical coating of copper surfaces by Ni Co Ni P Co P and Ni Co P alloys with catalytic and magnetic properties is described in detail The application of such layers is proposed PEMFC bipolar plates after additional coating by polytetrafluoroethylene The topic of the surface phosphatation is further extended by brief study of the properties of cesium dihydrogen phosphate for potential application in the fuel cell technology Meanwhile the Zn and Al anodization as metal surface treatment approach is also proposed Among the electrochemical analytical methods special attention is turned to Electrochemical Noise analysis Electrochemical Impedance Spectroscopy and to classical potentiodynamic scanning techniques *Nuclear Science Abstracts*, 1976-05

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