



ANALYTICAL CHEMISTRY INTERVIEW QUESTIONS AND ANSWERS GUIDE

- Q1.** What is the difference between qualitative and quantitative analysis?
- A1.** Qualitative analysis is the determination of the presence or absence of a particular element or compound in a sample. Quantitative analysis is the determination of the amount or concentration of a particular element or compound in a sample.
- Q2.** What is the difference between gravimetric and volumetric analysis?
- A2.** Gravimetric analysis is the determination of the amount of a substance by weighing it. Volumetric analysis is the determination of the amount of a substance by measuring the volume of a solution of known concentration that reacts with it.
- Q3.** What is the difference between titrimetric and colorimetric analysis?
- A3.** Titrimetric analysis is the determination of the amount of a substance by measuring the volume of a solution of known concentration that reacts with it. Colorimetric analysis is the determination of the amount of a substance by measuring the intensity of its color.
- Q4.** What is the difference between atomic absorption spectroscopy and atomic emission spectroscopy?
- A4.** Atomic absorption spectroscopy is the determination of the amount of a substance by measuring the absorption of light by its atoms. Atomic emission spectroscopy is the determination of the amount of a substance by measuring the emission of light by its atoms.
- Q5.** What is the difference between mass spectrometry and chromatography?
- A5.** Mass spectrometry is the determination of the mass-to-charge ratio of ions. Chromatography is the separation of a mixture into its individual components.
- Q6.** What is the difference between gas chromatography and liquid chromatography?
- A6.** Gas chromatography is the separation of a mixture into its individual components using a gas as the mobile phase. Liquid chromatography is the separation of a mixture into its individual components using a liquid as the mobile phase.
- Q7.** What is the difference between high performance liquid chromatography and gas chromatography?
- A7.** High performance liquid chromatography is a type of liquid chromatography that uses a high pressure pump to move the mobile phase through the column. Gas chromatography is a type of chromatography that uses a gas as the mobile phase.
- Q8.** What is the difference between thin layer chromatography and paper chromatography?
- A8.** Thin layer chromatography is a type of chromatography that uses a thin layer of adsorbent material on a plate. Paper chromatography is a type of chromatography that uses a piece of paper as the stationary phase.
- Q9.** What is the difference between infrared spectroscopy and ultraviolet-visible spectroscopy?
- A9.** Infrared spectroscopy is the determination of the amount of a substance by measuring the absorption of infrared light by its molecules. Ultraviolet-visible spectroscopy is the determination of the amount of a substance by measuring the absorption of ultraviolet or visible light by its molecules.
- Q10.** What is the difference between nuclear magnetic resonance spectroscopy and X-ray fluorescence spectroscopy?
- A10.** Nuclear magnetic resonance spectroscopy is the determination of the amount of a substance by measuring the absorption of radio waves by its nuclei. X-ray fluorescence spectroscopy is the determination of the amount of a substance by measuring the emission of X-rays by its atoms.

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