

Gravimetric Analysis of Chloride in Solution

Experiment 10
Lab 10
Date: _____

Introduction

This experiment involves the gravimetric analysis of a solution containing chloride ions. The procedure involves the precipitation of the chloride ions as silver chloride (AgCl) using a solution of silver nitrate (AgNO_3). The precipitate is then filtered, dried, and weighed. The mass of the precipitate is used to calculate the mass of chloride ions in the original solution. The experiment is performed in two parts. In the first part, a known mass of a sample containing chloride ions is dissolved in water. A known volume of a silver nitrate solution is added to the sample, and the mixture is allowed to react. The precipitate is then filtered, dried, and weighed. In the second part, a known mass of a sample containing chloride ions is dissolved in water. A known volume of a silver nitrate solution is added to the sample, and the mixture is allowed to react. The precipitate is then filtered, dried, and weighed. The mass of the precipitate is used to calculate the mass of chloride ions in the original solution.

The reaction between silver ions and chloride ions is as follows:

$$\text{Ag}^+ + \text{Cl}^- \rightarrow \text{AgCl} \downarrow$$

The precipitate is then filtered, dried, and weighed. The mass of the precipitate is used to calculate the mass of chloride ions in the original solution.

Procedure

The procedure for this experiment is as follows:

1. Weigh a known mass of a sample containing chloride ions.
2. Dissolve the sample in water.
3. Add a known volume of a silver nitrate solution to the sample.
4. Allow the mixture to react.
5. Filter the precipitate.
6. Dry the precipitate.
7. Weigh the precipitate.

Data Analysis

Sample 1	Sample 2
Mass of sample	Mass of sample
Mass of precipitate	Mass of precipitate
Mass of chloride ion	Mass of chloride ion

Calculate the mass of chloride ion in the sample 1.

$$\text{Mass of chloride ion} = \text{Mass of precipitate} \times \frac{\text{Molar mass of chloride ion}}{\text{Molar mass of silver chloride}}$$

Calculate the mass of chloride ion in the sample 2.

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Annelies Wilder-Smith



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web here is a list of a few things can you tell how long each of them can live exist a a dog d a human being b an elephant e a star f a mountain g a river c a tree 3 the poem

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web 7 answer the following questions a how does the brook sparkle b bicker means to quarrel why does the poet use this word here c how many hills and bridges does the brook pass during its journey d where does it finally meet the river e why has the word chatter been repeated in the poem f with many a curve my banks i

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web 3 who is i referred to in the poem the brook ans here i is referred to the brook in the poem brook because the poet made the brook as the speaker of the poem 4 what

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web the poem is about a brook a dictionary would define a brook as a stream or a small river read the poem silently first after the first reading the teacher will make you listen to a recording of the poem what do you think the poem is all about i come from haunts of coot and hern i make a sudden sally and sparkle out among the fern

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web the message of the poem is that the life of a brook is i temporary ii short lived iii eternal v momentary answer iii eternal question 6 b the poet draws a

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web the brook poem by lord alfred tennyson about the poem a read to understand 1 find the lines in the poem that express the following thoughts a as the brook flows

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web short answer type questions 30 40 words 2 marks each q 1 why has the poet used the word babble for the brook board term 12015 ans as the brook moves its

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web jun 30 2018 q1 who is i referred to as in the poem ans i in the poem is referred to as the brook although the writer of the poem is alfred tennyson but he has made

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web what is a stanza by stanza summary of the poem the brook by alfred lord tennyson what are the various words the poet uses to describe the sound of the brook how