IONIC & COVALENT BONDS

Name: Date: Period:

An _____ is the attraction between two oppositely charged ions. The chemical bond formed when two atoms share electrons is called a _____. Covalent bonds usually form between atoms of ____ and ionic bonds usually form when a ____.

Activity A: How Do Ions Form?

Procedure:

- Place three pairs of checkers (three red and three black) on your desk. The red represent electrons and the black represent protons.
- Place nine pairs of checkers (nine red and nine black) in a separate group on your desk.
- 3. Move a red checker from the smaller group to the larger group.
- 4. Count the number of positive charges (protons) and negative charges (electrons) in each group.
- Now sort the checkers into a group of four pairs and a group of eight pairs. Repeat Steps 3 and 4, this time moving two red checkers form the smaller group to the larger group.

Think it over

- 1. What was the total charge on each group before you moved the red checkers (electrons)?
- 2. What was the charge on each group after you moved the checkers?
- 3. Based on this activity what do you think happens to the charge on an atom when it loses electrons?
- 4. Based on this activity what do you think happens to the charge on an atom when it gains electrons?

Activity B: Interpreting Data About Ions

Procedure:

- Use the periodic table and the chart to the right to identify the charges of the ions in each ionic compound listed below.
- 2. Write the formulas for each compound.
 - a. sodium beomide:
 - b. lithium oxide:
 - c. magnesium sulfide:
 - d. aluminum fluoride:
 - e. potassium nitrate:
 - ammonium chloride

Ions and Their Charges		
Name	Charge	Symbol or Formula
Lithium	1+	Li*
Sodium	1.+	Na"
Potassium	1+	K*
Ammonium	1+	NH ₄ *
Calcium	2+	Ca ² *
Magnesium	2+	Mg ²⁺
Aluminum	3+	Al ³ *
Floride	1-	F.
Chloride:	1-	CT
Iodide	11-	
Bicarbonate	11-	HCO ₃
Nitrate	1-	NO ₃
Oxide	2-	O ²
Sulfide	2-	S2-
Carbonate	2-	CO ₃ 2.
Sulfate	2-	SO_4^{-2}
Phosphate	3-	PO ₄ ³

Think it over:

1. How do you know how many of each ion to write in the formula?

Answers Covalent Bonding Vocabulary

Victor M. Corman

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Master Vocabulary Workbook David Nicholls, 2023-06-11 Supplementary Exercises for my Master English Vocabulary course available on skype lessons com and udemy com Science Vocabulary Building, Grades 5 - 8 Schyrlet Cameron, Carolyn Craig, 2009-02-16 Connect students in grades 5 8 with science using Science Vocabulary Building This 80 page book reinforces commonly used science words builds science vocabulary and increases students readability levels This comprehensive classroom supplement includes alphabetized word lists that provide pronunciations syllabications definitions and context sentences for high utility science words Activities allow for differentiated instruction and can be used as warm ups homework assignments and extra practice The book supports National Science Education Standards Science to English Learners Stephen Fleenor, Tina Beene, 2019 Teaching Science to English Learners is a go to resource for science educators to promote listening speaking reading and writing in their classrooms Each chapter outlines specific easy to implement strategies that foster academic language development and comprehension of science concepts with specific scaffolds highlighted for each language proficiency level Teaching Science to English Learners is thoughtfully designed to help teachers equip students to navigate inquiry driven instruction understand abstract science concepts and master the multitude of science vocabulary that can be challenging for ELs and non ELs alike By equipping teachers with strategies to draw from students prior knowledge and focus on collaboration this book helps science teachers make lessons accessible for all learners while deepening content comprehension and developing academic language Teaching Science to English Learners is the second in a set of books designed to help content area teachers create learning environments in which English learners thrive E3 Chemistry Review Book - 2018 Home Edition (Answer Key Included) Effiong Eyo, 2017-10-20 With Answer Key to All Questions Chemistry students and homeschoolers Go beyond just passing Enhance your understanding of chemistry and get higher marks on homework guizzes tests and the regents exam with E3 Chemistry Review Book 2018 With E3 Chemistry Review Book students will get clean clear engaging exciting and easy to understand high school chemistry concepts with emphasis on New York State Regents Chemistry the Physical Setting Easy to read format to help students easily remember key and must know chemistry materials Several example problems with solutions to study and follow Several practice multiple choice and short answer questions at the end of each lesson to test understanding of the materials 12 topics of Regents question sets and 3 most recent Regents exams to practice and prep for any Regents Exam This is the Home Edition of the book Also available in School Edition ISBN 978 197836229 The Home Edition contains an answer key section Teachers who want to recommend our Review Book to their students should recommend the Home Edition Students and and parents whose school is not using the Review Book as instructional material as well as homeschoolers should buy the Home Edition The School Edition does not have answer key in the book A separate answer key booklet is provided to teachers with a class order of the book Whether you are using the school or Home Edition our E3

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