
Chem - Covalent Bonding: Questions & Answers

properties of non metals ✓ Ans - inability to conduct electricity - absence of ions or delocalised electrons
low boiling and melting points - generally weak intermolecular bonds, can be soft

molecules ✓ Ans - separate groups of atoms with a formula, bonded together

intermolecular bonds ✓ Ans - bonds between molecules (dispersion, dipole-dipole, hydrogen bonding) determines physical properties

intramolecular bonds ✓ Ans - bonds between atoms in a molecule (covalent, ionic eg.)

covalent bonds ✓ Ans - non metals share electrons to complete the octet rule, an atom forms as many bonds as it needs eg. carbon needs 4 electrons so forms 4 bonds

why do non metals share electrons not transfer? ✓ Ans - non metals have a relatively high electronegativity so competing to attract each others electron, shares instead

electron dot diagrams ✓ Ans - lewis diagram shows valence electrons (bonding pairs) and lone pairs

valence structure ✓ Ans - each pair of bonding electrons is shown as a line, still represents dots as lone pairs. 1 line=2 electrons

bonding pairs and lone pairs ✓ Ans - bonding pairs; shared electrons in bond
lone pairs; non bonding pairs

double covalent bond ✓ Ans - rigid, two pairs of electrons (4 electrons) are shared, represented as a double line in valence structures eg. O_2

Answers For Chem Think Covalent Bonding

LP Steffe



Answers For Chem Think Covalent Bonding:

Chemistry of the Covalent Bond Leallyn Burr Clapp, 1957

Whispering the Techniques of Language: An Mental Quest through **Answers For Chem Think Covalent Bonding**

In a digitally-driven earth wherever screens reign great and instant conversation drowns out the subtleties of language, the profound techniques and psychological nuances hidden within phrases frequently get unheard. However, located within the pages of **Answers For Chem Think Covalent Bonding** a captivating fictional prize blinking with organic feelings, lies a fantastic journey waiting to be undertaken. Written by an experienced wordsmith, that wonderful opus attracts readers on an introspective trip, lightly unraveling the veiled truths and profound affect resonating within the fabric of each word. Within the psychological depths of the moving review, we can embark upon a genuine exploration of the book is primary subjects, dissect its fascinating publishing style, and fail to the powerful resonance it evokes deep within the recesses of readers hearts.

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