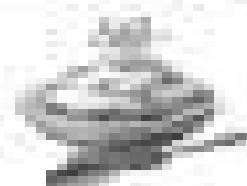


Atomic Probing Structure of the Atom

Name _____
Date _____



Purpose:

To determine the structure of the atom by observing the deflection of alpha particles.

Part I: Understanding the Structure of the Atom

The Rutherford alpha particle scattering experiment was a key experiment in the development of the atomic model. It was designed to determine the structure of the atom by observing the deflection of alpha particles. The results of the experiment showed that the atom is mostly empty space, with a small, dense, positively charged nucleus at the center.

Objective: To determine the structure of the atom by observing the deflection of alpha particles.

The Rutherford alpha particle scattering experiment was a key experiment in the development of the atomic model. It was designed to determine the structure of the atom by observing the deflection of alpha particles.

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Answers Lesson 11 Atomic Pudding

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