#### AP Bio Chapter 06 Group Notes: Metabolism - Energy and Enzymes

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Team Members:

HSR&T (Highly Suggested Reading and Thinking) This is independent work that you can do as a (GSN) team/individually - either way, ALL members of the team should do the reading/thinking....you will be assessed over this information.

Big Idea 4: Interdependent Relationships ~ "Enzyme Inhibitors Can Spell Death" (p. 108)

- How do the following specifically impact organisms? —also where have they been discovered or used by people?
  - a. Cyanide gas/MPTP;
  - b. Sarin:
  - c. Wartarin:
- 2. Give an example of how our understanding of warfarin has had positive consequences:

#### 6.1 Cells and the Flow of Energy 3.4

All end	y: the ability to do -work- or bring about a change ergy - dependent upon the sun ultimately! Kinetic energy: anything -moving- has it Potential energy: Stored energy Chemical energy: form of potential energy that is stored in -molecules dependent upon its -bonds-	
	of Thermodynamics:	Autotrophs= Primary producers that can
1.	Basics: the study of energy transformations that	create organic energy using non-organic
-	occur in matter	materials like carbon-dioxide or solar energy
2.	a. also called: -Law of Conservation of Energy- b. states: energy cannot be created or destroyed, but it can be transferred or transformed. c. example: solar energy used to form carbohydrates by autotrophs	(Plants are a form of autotrophs)
1.3	Second Law	
	a. states: every energy transfer or transformation increases the entropy, or the amount of —disorder/randomness— in the universe     b. example: heat generated by you working out, no transfer of energy is 100% efficient.	
4.	Summary: Can't create energy - only -convert-, at every conversion, some of the energy will dissipate as -heat- and become less useful.	
Entropy:		Heat = Increased entropy in the universe
	Defined: Relative amount of -disorganization-	(increased amount of disorganization)
2.	Examples: Glucose breaking down into carbon dioxide and water	
3.	Summary:	
	<ul> <li>energy conversions result in heat,</li> <li>therefore, the entropy of the universe is</li> </ul>	

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**S Marginson** 

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