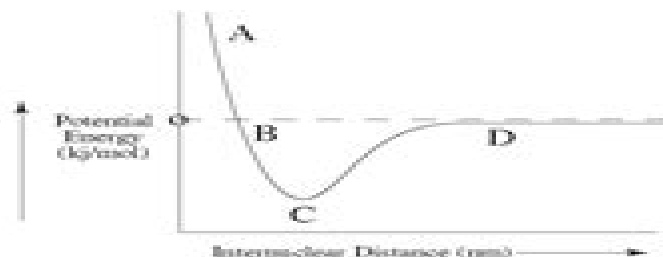




The above reaction will experience a rate increase by the addition of a catalyst such as platinum. Which of the following best explains why?

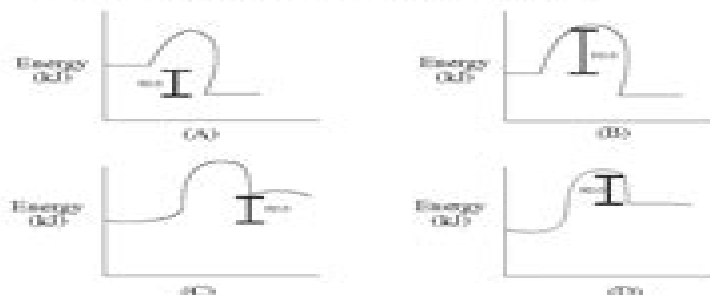
- (A) The catalyst increases the overall frequency of collisions in the reactant molecules.
 - (B) The catalyst increases the frequency of collisions that occur at the proper orientation in the reactant molecules.
 - (C) The catalyst introduces a new reaction mechanism for the reaction.
 - (D) The catalyst increases the activation energy for the reaction.
57. The graph below shows the amount of potential energy between two hydrogen atoms as the distance between them changes. At which point in the graph would a molecule of H_2 be the most stable?



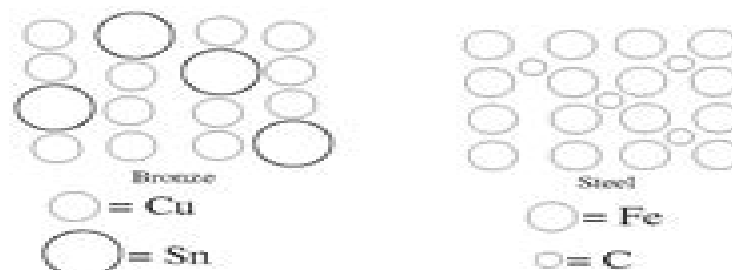
- (A) Point A
 - (B) Point B
 - (C) Point C
 - (D) Point D
58. $\text{N}_2(g) + \text{O}_2(g) + \text{Cl}_2(g) \rightleftharpoons 2\text{NOCl}(g) \quad \Delta G^\circ = 132.6 \text{ kJ/mol}$
- For the equilibrium above, what would happen to the value of ΔG if the concentration of N_2 were to increase and why?
- (A) It would increase as the reaction would become more thermodynamically favored.
 - (B) It would increase as the reaction would shift right and create more products.
 - (C) It would decrease because there are more reactants present.
 - (D) It would stay the same because the value of K_{eq} would not change.



Which of the following energy level diagrams gives an accurate representation of the above reaction?



60. Two alloys are shown in the diagrams above—bronze, and steel. Which of the following correctly describes the malleability of both alloys compared to their primary metals?



- (A) Bronze's malleability would be comparable to that of copper, but steel's malleability would be significantly lower than that of iron.
- (B) Bronze's malleability would be significantly higher than that of copper, but steel's malleability would be comparable to that of iron.
- (C) Both bronze and steel would have malleability values similar to those of their primary metals.
- (D) Both bronze and steel would have malleability values greater than those of their primary metals.

END OF SECTION I

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