

Worksheet #3: Naming Inorganic Compounds

1. Give the name for each of the following ionic compounds.

- AlF_3 Aluminum Fluoride
- $\text{Fe}(\text{OH})_3$ Iron(III) Hydroxide
- $\text{Cu}(\text{NO}_3)_2$ Copper (II) Nitrate
- $\text{Ba}(\text{ClO}_4)_2$ Barium Perchlorate
- LiNO_3 Lithium Nitrate
- Hg_2S Mercury (I) Sulfide
- $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$ Calcium Acetate
- $\text{Cr}(\text{CO}_3)_3$ Chromium (III) Carbonate
- $\text{K}_2\text{Cr}_2\text{O}_7$ Potassium Dichromate
- $\text{Pb}(\text{NO}_3)_2$ Lead(II) Nitrate

(2) all of the underlined compounds are Polyatomic Ions $\rightarrow$ Memorise These

2. Write the chemical formula for each of the following compounds:

- a. copper(II) oxide Cu_2O
- b. potassium peroxide K_2O_2
- c. aluminum hydroxide $\text{Al}(\text{OH})_3$
- d. zinc sulfate $\text{Zn}(\text{SO}_4)$
- Exerc. 15. a. titanium(III) oxide, Ti_2O_3

Figure 13. 2010-2011 Season: Catches of 1

- a. mercury(I) bromide Hg_2Br_2
- b. iron(III) carbonate $\text{Fe}_2(\text{CO}_3)_3$
- c. sodium hypobromite NaBrO not
- most common polyatomic ions
 BrO^{1-}

$\text{R-O}^{\ominus} \text{R} + \text{H}^{\oplus} \text{O} \rightarrow \text{R-OH} + \text{R-O}^{\ominus} \text{H}^{\oplus}$

2. Give the name or chemical formula, as appropriate, for each of the following acids:

- HNO_3 Nitric Acid
- HNO_2 Hydro Nitric Acid
- H_3PO_4 Phosphoric Acid
phosphate (the -ate)

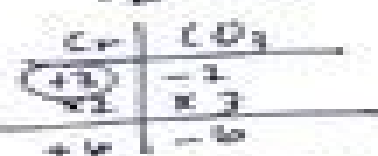
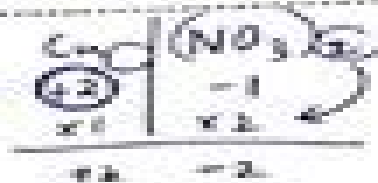
- d. hypochlorous acid HClO
e. iodic acid HIO_3
f. sulfurous acid H_2SO_3
→ break off of iodate as IO_3^-

^a Based off of Totality TO₂

4. Give the name or chemical formula, as appropriate, for each of the following molecular substances:

1. Iron(III) hexafluoride
2. Iron(III) hexafluoride
3. Iron(III) hexafluoride

- d. dinitrogen tetroxide: N_2O_4
e. hydrogen cyanide: HCN
f. tetraphosphorus hexasulfide: P_4S_6



— *W. J. G. B. J.*



1. The first step is to identify the problem.
 2. The second step is to define the problem.
 3. The third step is to analyze the problem.
 4. The fourth step is to develop a solution.
 5. The fifth step is to implement the solution.
 6. The sixth step is to evaluate the solution.
 7. The seventh step is to monitor the solution.
 8. The eighth step is to maintain the solution.
 9. The ninth step is to improve the solution.
 10. The tenth step is to document the solution.



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Table of Contents Ap Chemistry Summer Packet Answers

1. Understanding the eBook Ap Chemistry Summer Packet Answers
 - The Rise of Digital Reading Ap Chemistry Summer Packet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ap Chemistry Summer Packet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ap Chemistry Summer Packet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ap Chemistry Summer Packet Answers
 - Personalized Recommendations
 - Ap Chemistry Summer Packet Answers User Reviews and Ratings
 - Ap Chemistry Summer Packet Answers and Bestseller Lists
5. Accessing Ap Chemistry Summer Packet Answers Free and Paid eBooks
 - Ap Chemistry Summer Packet Answers Public Domain eBooks
 - Ap Chemistry Summer Packet Answers eBook Subscription Services
 - Ap Chemistry Summer Packet Answers Budget-Friendly Options

6. Navigating Ap Chemistry Summer Packet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ap Chemistry Summer Packet Answers Compatibility with Devices
 - Ap Chemistry Summer Packet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ap Chemistry Summer Packet Answers
 - Highlighting and Note-Taking Ap Chemistry Summer Packet Answers
 - Interactive Elements Ap Chemistry Summer Packet Answers
8. Staying Engaged with Ap Chemistry Summer Packet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ap Chemistry Summer Packet Answers
9. Balancing eBooks and Physical Books Ap Chemistry Summer Packet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ap Chemistry Summer Packet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ap Chemistry Summer Packet Answers
 - Setting Reading Goals Ap Chemistry Summer Packet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ap Chemistry Summer Packet Answers
 - Fact-Checking eBook Content of Ap Chemistry Summer Packet Answers
 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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