#### Balancing Chemical Equations Worksheet

1. 
$$2 H_2 + 0_2 \rightarrow 2 H_2O$$

4. 
$$2 N_2 + 0_2 \rightarrow 2 N_20$$

7. 
$$Zn + Z HCI \rightarrow ZnCl_2 + H_2$$

8. \_\_\_\_ SiCl<sub>4</sub> + \_\_\_ H<sub>2</sub>O 
$$\Rightarrow$$
 \_\_\_\_ H<sub>4</sub>SiO<sub>4</sub> + \_\_ HCl

9. 
$$2 \text{ Na} + 2 \text{ H}_2\text{O} \rightarrow 2 \text{ NaOH} + H_2$$

10. 
$$_{-2}$$
  $H_3PO_4 \rightarrow _{---}$   $H_4P_2O_7 + _{---}$   $H_2O$ 

12. \_\_\_\_ 
$$CO_2$$
 + \_\_\_  $Z$   $NH_3  $\rightarrow$  ____  $OC(NH_2)_2$  + ____  $H_2O$$ 

13. 
$$4 \text{ Si}_2\text{H}_3 + 7 \text{ O}_2 \rightarrow 8 \text{ SiO}_2 + 6 \text{ H}_2\text{O}_3$$

14. 
$$2 \text{ Al}(OH)_3 + 3 \text{ H}_2SO_4 \rightarrow Al_2(SO_4)_3 + 6 \text{ H}_2O$$

16. \_\_\_\_ Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> + \_\_\_\_\_\_ KOH 
$$\rightarrow$$
 \_\_\_\_\_\_ K<sub>2</sub>SO<sub>4</sub> + \_\_\_\_\_ Fe(OH)<sub>3</sub>

17. 
$$2 C_7H_6O_2 + 15 O_2 \rightarrow 14 CO_2 + 5 H_2O$$

19. \_\_\_\_ FeS<sub>2</sub> + \_\_\_\_ O<sub>2</sub> 
$$\rightarrow$$
 \_\_\_\_ Fe<sub>2</sub>O<sub>3</sub> + \_\_\_\_ SO<sub>2</sub>

22. \_\_\_\_ Na<sub>2</sub>CO<sub>3</sub> + \_\_2\_ HCl 
$$\rightarrow$$
 \_2\_ NaCl + \_\_\_\_ H<sub>2</sub>O + \_\_\_ CO<sub>2</sub>

24. \_\_\_\_ 
$$C_7H_{16} + _//_ O_2 \rightarrow _//_ CO_2 + _//__ H_{2}O$$

25.  $P_4 + 5 O_2 \rightarrow 2 P_2O_5$ 

# **Answers For Balancing Equations**

Puja Mehta

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