

Question number	Answer	Marks	Guidance															
1 (a)	C	B1																
1 (b)	B	B1																
1 (c)	D	B1																
2 (a)	38	B1																
2 (b)	32	B1																
2 (c)	5	B1																
3 (a) (i)	23	B1																
3 (a) (ii)	$^{54}_{26}\text{Fe}$	B2																
3 (b)	$^{23}_{11}\text{Na}$	B2																
3 (c)	$^{88}_{38}\text{Sr}$	B2																
4 (a)	Isotopes are atoms of the same element with different numbers of neutrons and different masses.	B1																
4 (b)	Relative isotopic mass is the mass of an isotope relative to one-twelfth of the mass of an atom of carbon-12.	B2																
4 (c)	71	B1																
4 (d)	<table><tr><td></td><td>protons</td><td>neutrons</td><td>electrons</td><td>charge</td></tr><tr><td>^{17}O</td><td>8</td><td>9</td><td>10</td><td>2-</td></tr><tr><td>^{27}Al</td><td>13</td><td>14</td><td>10</td><td>3+</td></tr></table>		protons	neutrons	electrons	charge	^{17}O	8	9	10	2-	^{27}Al	13	14	10	3+	B2	
	protons	neutrons	electrons	charge														
^{17}O	8	9	10	2-														
^{27}Al	13	14	10	3+														
5 (a)	Relative atomic mass is the weighted mean mass of an atom of an element relative to one-twelfth of the mass of an atom of carbon-12.	B3																
5 (b)	$A_r = 32.09$	B2																
5 (c)	<table><tr><td></td><td>protons</td><td>neutrons</td><td>electrons</td></tr><tr><td>^{32}S</td><td>16</td><td>16</td><td>16</td></tr><tr><td>$^{32}\text{S}^{2-}$</td><td>16</td><td>16</td><td>18</td></tr></table>		protons	neutrons	electrons	^{32}S	16	16	16	$^{32}\text{S}^{2-}$	16	16	18	B1				
	protons	neutrons	electrons															
^{32}S	16	16	16															
$^{32}\text{S}^{2-}$	16	16	18															
6 (a)	$3\text{Mg(s)} + \text{N}_2\text{(g)} \rightarrow \text{Mg}_3\text{N}_2\text{(s)}$ 1 mark for correct species and state symbols 1 mark for balance	B1 × 2																
6 (b)	$\text{Ca(s)} + 2\text{H}_2\text{O(l)} \rightarrow \text{Ca(OH)}_2\text{(aq)} + \text{H}_2\text{(g)}$ 1 mark for correct species and state symbols 1 mark for balance	B1 × 2																
6 (c)	$6\text{NaOH(aq)} + \text{Fe}_2(\text{SO}_4)_3\text{(aq)} \rightarrow 2\text{Fe(OH)}_3\text{(s)} + 3\text{Na}_2\text{SO}_4\text{(aq)}$ 1 mark for correct species and state symbols 1 mark for balance	B1 × 2																

Answers To Unit 2 As Level Chemistry

Judith Potter, Peter Cann



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McFarland is a senior examiner and is Head of Chemistry at Regent House Grammar School He has taught CCEA Chemistry at all levels for 20 years His previous publications include the CCEA GCSE Single Award Science Foundation Tier textbook published by Hodder Education **CCEA A2 Unit 2 Chemistry Student Guide: Analytical, Transition Metals,**

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point slope is the general form $y - y_1 = m(x - x_1)$ for linear equations it emphasizes the slope of the line and a point on the line that is not the y intercept we can rewrite an equation in point slope form to be in slope intercept form $y = mx + b$ to highlight the same line s slope and y intercept

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examples example 1 find an equation of the line with a slope of 3 that passes through the point $(2, 4)$ the slope of 3 tells us to replace the m with 3 the point $(2, 4)$ tells us that $x - x_1$

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overview of different forms of a line's equation there are many different ways that you can express the equation of a line there is the slope intercept form standard form and also this page's topic point slope form each one expresses the equation of a line and each one has its own pros and cons point slope form this page's topic makes it easy to find the line's

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the point slope form is very useful when you don't have your y intercept it is used to write equations when you only have your slope and a point point slope form $y - y_1 = m(x - x_1)$ for example your slope m is 3 and your point a, b is $(9, 10)$ you would substitute your y coordinate for a and your x coordinate for b

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structure $y - y_1 = m(x - x_1)$ where m is the slope of the line x_1, y_1 is a point on the line and x and y are variables representing other points on the line

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point slope form in point slope form which is written like this $y - y_1 = m(x - x_1)$ y_1 is the y value of the known point on the line m is the slope and x_1 is the x value of the

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