

Notes

1. How are different types of radiation arranged along the electromagnetic spectrum?

- By how fast they travel
- By their sources
- By the amount of energy they carry
- By how radioactive they are

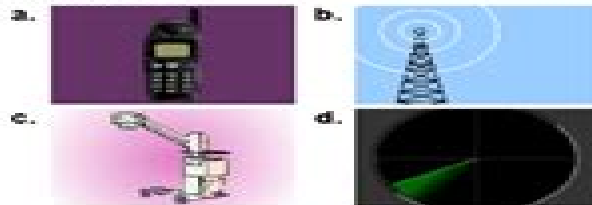
2. What is the difference between a wave and a particle?

- A particle is a discrete unit of energy; a wave is a continuous flow of energy
- A wave carries energy; a particle carries light
- A wave moves at the speed of light; a particle can move at a range of speeds
- A wave carries electricity; a particle carries a magnetic charge

3. Place the following types of EM radiation in order, from lowest energy to highest energy: A) Visible light; B) Gamma rays; C) Radio waves

- A, C, B
- C, A, B
- B, C, A
- C, B, A

4. Which of the following transmits information using microwaves?



5. Astronomers often scan the sky using devices called radio telescopes. What can you conclude about these devices?

- They send radio broadcasts out to distant stars and planets.
- They pick up the radio waves emitted by stars.
- They are less powerful than optical telescopes.
- They can detect ultraviolet radiation.

Name: _____
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Tim mentions that infrared radiation can be used to take special pictures called thermographs. "Graph" means "image." What can you infer about the prefix "thermo?"

- It means "light"
- It means "energy"
- It means "heat"
- It means "radio"



What might happen if the ozone layer did not block the sun's ultraviolet rays?

- Humans would be able to see in the ultraviolet range of the EM spectrum
- Devices that use electromagnetic radiation would no longer work
- The surface of the earth would be much hotter
- Rates of skin cancer would skyrocket

8. Which word best describes gamma rays?

- Weak
- Destructive
- Visible
- Harmless

9. What do all forms of electromagnetic radiation have in common?

- Their strength
- Their speed
- Their wavelength
- Their frequency

10. What produces electromagnetic radiation?

- Electrons changing energy levels
- Electrons bonding with other atoms
- Electrons leaving the atomic nucleus
- Electrons becoming protons or neutrons

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