

## AP Chemistry: Gases

### Multiple Choice

(You may use a calculator.)

21. When a sample of oxygen gas in a closed container of constant volume is heated until its absolute temperature is doubled, which of the following is also doubled?

- (A) The density of the gas                      (B) The pressure of the gas                      (C) The average velocity of the gas molecules  
(D) The number of molecules per  $\text{cm}^3$                       (E) The potential energy of the molecules

23. The density of an unknown gas is 4.20 grams per liter at 3.00 atmospheres pressure and 127 °C. What is the molecular weight of this gas? ( $R = 0.0821 \text{ L}\cdot\text{atm} / \text{mole}\cdot\text{K}$ )

- (A) 14.6                      (B) 46.0                      (C) 88.0                      (D) 94.1                      (E) 138

39. Equal masses of three different ideal gases, X, Y, and Z, are mixed in a sealed rigid container. If the temperature of the system remains constant, which of the following statements about the partial pressure of gas X is correct?

- (A) It is equal to 1/3 the total pressure.  
(B) It depends on the intermolecular forces of attraction between molecules of X, Y, and Z.  
(C) It depends on the relative molecular masses of X, Y, and Z.  
(D) It depends on the average distance traveled between molecular collisions.  
(E) It can be calculated with knowledge only of the volume of the container.

50. Two flexible containers for gases are at the same temperature and pressure. One holds 0.50 grams of hydrogen and the other holds 8.0 grams of oxygen. Which of the following statements regarding these gas samples is FALSE?

- (A) The volume of the hydrogen container is the same as the volume of the oxygen container.  
(B) The number of molecules in the hydrogen container is the same as the number of molecules in the oxygen container.  
(C) The density of the hydrogen sample is less than that of the oxygen sample.  
(D) The average kinetic energy of the hydrogen molecules is the same as the average kinetic energy of the oxygen molecules.  
(E) The average speed of the hydrogen molecules is the same as the average speed of the oxygen molecules.

52.  $3 \text{ Ag}_{(s)} + 4 \text{ HNO}_{3(aq)} \rightleftharpoons 3 \text{ AgNO}_{3(aq)} + \text{NO}_{(g)} + 2 \text{ H}_2\text{O}_{(l)}$

The reaction of silver metal and dilute nitric acid proceeds according to the equation above. If 0.10 moles of powdered silver is added to 10.0 milliliters of 6.0-molar nitric acid, the number of moles of NO gas that can be formed is...

- (A) 0.015 mole                      (B) 0.020 mole                      (C) 0.030 mole                      (D) 0.045 mole                      (E) 0.090 mole

78. When the actual gas volume is greater than the volume predicted by the ideal gas law, the explanation lies in the fact that the ideal gas law does NOT include a factor for molecular...

- (A) volume                      (B) mass                      (C) velocity                      (D) attractions                      (E) shape

72. A compound is heated to produce a gas whose molecular weight is to be determined. The gas is collected

# Ap Chemistry Chapter Gases Law Quiz

**H Kauffman**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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