#### Balancing Chemical Equations Worksheet

1. 
$$2 H_2 + 0_2 \rightarrow 2 H_2O$$

2. 
$$N_2 + 3 H_2 \rightarrow 2 NH_3$$

4. 
$$2 N_2 + 0_2 \rightarrow 2 N_20$$

5. 
$$2 \text{ HgO} \rightarrow 2 \text{ Hg} + 0$$

6. 
$$6 CO_2 + 6 H_2O \rightarrow C_6H_{12}O_6 + 6 O_2$$

9. 
$$\underline{2}$$
 Na +  $\underline{2}$  H<sub>2</sub>O  $\Rightarrow$   $\underline{2}$  NaOH +  $\underline{\phantom{0}}$  H<sub>2</sub>

10. 
$$_{-2}$$
  $H_3PO_4 \rightarrow _{---}$   $H_4P_2O_7 + _{---}$   $H_2O$ 

12. \_\_\_\_ 
$$CO_2$$
 + \_\_\_  $Z$   $NH_3  $\rightarrow$  ____  $OC(NH_2)_2$  + ____  $H_2O$$ 

13. 
$$4 \text{ Si}_2\text{H}_3 + 7 \text{ O}_2 \rightarrow 8 \text{ SiO}_2 + 6 \text{ H}_2\text{O}_3$$

14. \_2\_Al(OH)<sub>3</sub> + \_3\_H<sub>2</sub>SO<sub>4</sub> 
$$\Rightarrow$$
 \_\_\_Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> + \_6\_H<sub>2</sub>O

16. \_\_\_\_ Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> + \_\_\_6\_ KOH 
$$\rightarrow$$
 \_\_\_3\_ K<sub>2</sub>SO<sub>4</sub> + \_\_2\_ Fe(OH)<sub>3</sub>

17. 
$$2 C_7H_6O_2 + 15 O_2 \rightarrow 14 CO_2 + 5 H_2O$$

19. \_\_\_\_\_ FeS<sub>2</sub> + \_\_\_\_\_ O<sub>2</sub> 
$$\rightarrow$$
 \_\_\_\_\_ Fe<sub>2</sub>O<sub>3</sub> + \_\_\_\_\_ SO<sub>2</sub>

21. \_\_\_\_Fe<sub>2</sub>O<sub>3</sub> + \_\_3\_H<sub>2</sub> 
$$\rightarrow$$
 \_\_2\_Fe + \_\_3\_H<sub>2</sub>O

22. \_\_\_\_Na<sub>2</sub>CO<sub>3</sub> + \_\_\_ HCl 
$$\rightarrow$$
 \_\_ NaCl + \_\_\_\_H<sub>2</sub>O + \_\_\_\_CO<sub>2</sub>

24. \_\_\_\_ 
$$C_7H_{16} + _//_ O_2 \rightarrow _//_ CO_2 + _//__ H_{2}O$$

25. 
$$P_4 + 5 O_2 \rightarrow 7 P_2O_5$$

# **Answers For Balancing Equations**

H. Stephen Stoker, Danny V. White

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## **Table of Contents Answers For Balancing Equations**

- 1. Understanding the eBook Answers For Balancing Equations
  - The Rise of Digital Reading Answers For Balancing Equations
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Answers For Balancing Equations
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Answers For Balancing Equations
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Answers For Balancing Equations
  - Personalized Recommendations
  - Answers For Balancing Equations User Reviews and Ratings
  - Answers For Balancing Equations and Bestseller Lists
- 5. Accessing Answers For Balancing Equations Free and Paid eBooks
  - Answers For Balancing Equations Public Domain eBooks
  - Answers For Balancing Equations eBook Subscription Services
  - Answers For Balancing Equations Budget-Friendly Options

- 6. Navigating Answers For Balancing Equations eBook Formats
  - o ePub, PDF, MOBI, and More
  - Answers For Balancing Equations Compatibility with Devices
  - Answers For Balancing Equations Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Answers For Balancing Equations
  - Highlighting and Note-Taking Answers For Balancing Equations
  - Interactive Elements Answers For Balancing Equations
- 8. Staying Engaged with Answers For Balancing Equations
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Answers For Balancing Equations
- 9. Balancing eBooks and Physical Books Answers For Balancing Equations
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Answers For Balancing Equations
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Answers For Balancing Equations
  - Setting Reading Goals Answers For Balancing Equations
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers For Balancing Equations
  - Fact-Checking eBook Content of Answers For Balancing Equations
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

• Interactive and Gamified eBooks

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