



Parent Answer Key

Grade 6 Math & Reading

Companion Work Text & Student Practice Book

At-Home Learning

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Also included are directions for helping your student use the Student Answer Form and for you to quickly grade your student's work to the multiple-choice questions using the foldable answer key and answer form.

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Judd E. Hollander



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web there are two main rules to remember when balancing nuclear equations 1 the total of the superscripts mass numbers in the reactants and products must be the same 2 the total of the subscripts atomic numbers in the reactants and products must be the same p a r t a what is the value of in the following nuclear reaction

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web we can write a balanced nuclear equation for this reaction as let's do a practice example identify the product x by writing a balanced nuclear equation $137\text{ Cs} \rightarrow 137\text{ Ba} + x$ because the mass number is the same on both sides of the equation x must have a mass number of 0

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web nuclear reactions are going on all around us using correctly balanced equations is important when trying to understand nuclear reactions all equations need to be balanced to conform to two conservation laws the mass number is conserved and the electrical charge of the nucleus is conserved