



NAME: Abhishek H. Nair Div: A Roll No. 1

SCIENCE Paper – 2*(Two hours)**You will not be allowed to write during the first 15 minutes.**This time is to be spent in reading the Question Paper.**The time given at the head of this paper is the time allowed for writing the answers.**Section I is compulsory. Attempt any four questions from Section II.**The intended marks for questions or parts of questions are given in brackets []***SECTION I (40 Marks)***(Attempt ALL questions)***Question 1****[15]**

Choose the correct answer from the options given below:

(Do not copy the question; write the correct answers only.)

- Whenever the gases react under similar conditions of temperature and pressure, they do so in volumes which bear a simple ratio of each other and the products if gaseous. The above law was stated by
 a) Boyle b) Charles c) Gay Lussac d) Avogadro
- A strip of copper is placed in four different colourless salt solutions. After electrolysis, one of the salt solutions becomes blue in colour. Identify the salt solution which turns blue from the given salt solutions:
 a) K_2SO_4 b) $CuSO_4$ c) $AgSO_4$ d) $CaSO_4$
- Potassium is the element below sodium in the long form of the periodic table. The electronegativity of potassium will be
 a) Lower than sodium b) Higher than sodium c) Same as sodium d) None of these
- The salt solution which does not react with ammonium hydroxide is :
 a) Calcium Nitrate b) Zinc Nitrate c) Lead Nitrate d) Copper Nitrate
- Assertion :** Sodium is the most reactive metal among group 1 elements.
Reason : In case of metals, greater the tendency to lose electrons, greater is the reactivity.
 a) A and R both are correct statements and R is the correct explanation of the A
 b) A and R both are correct statements but R is not the correct explanation of the A
 c) A is true, but R is false
 d) A is false , but R is true



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Answers For A Ls Chemistry

Charles E. Mortimer



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decide in which line you want to study and either speak to your employer or contact your nearest labour centre or provincial

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combined with practical on the job experience we learn much better when we can practice what we have been taught in the

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