

Practicing the Mole - - Even Problems Answer Key

Calculate the mass in grams of each of the following:

2. 8.00 moles of aluminum

$$\frac{8.00 \text{ mol} \mid 26.98 \text{ g}}{1 \text{ mol}} = 216 \text{ g Al}$$

6. 7.00 moles of iodine (I_2)

$$\frac{7.00 \text{ mol} \mid 253.8 \text{ g}}{1 \text{ mol}} = 1,780 \text{ g I}_2$$

4. 2.00×10^2 moles of chlorine (Cl_2)

$$\frac{200. \text{ mol} \mid 70.90 \text{ g}}{1 \text{ mol}} = 14,200 \text{ g Cl}_2$$

8. 9.20 moles of iron

$$\frac{9.20 \text{ mol} \mid 55.85 \text{ g}}{1 \text{ mol}} = 514 \text{ g Fe}$$

Calculate the number of moles of atoms in each of the following:

10. 281 g of iron

$$\frac{281 \text{ g} \mid 1 \text{ mol}}{55.85 \text{ g}} = 5.03 \text{ mol Fe}$$

14. 93.0 g of phosphorus

$$\frac{93.0 \text{ g} \mid 1 \text{ mol}}{30.97 \text{ g}} = 3.00 \text{ mol P}$$

12. 72.0 g of magnesium

$$\frac{72.0 \text{ g} \mid 1 \text{ mol}}{24.31 \text{ g}} = 2.96 \text{ mol Mg}$$

16. 1035 g of lead

$$\frac{1035 \text{ g} \mid 1 \text{ mol}}{207.2 \text{ g}} = 4.995 \text{ mol Pb}$$

Calculate the number of atoms in each of the following:

18. 5.40 moles of iron

$$\frac{5.40 \text{ mol} \mid 6.022 \times 10^{23} \text{ atoms}}{1 \text{ mole}} = 3.25 \times 10^{24} \text{ atoms Fe}$$

20. 10.0 moles of titanium

$$\frac{10.0 \text{ mol} \mid 6.022 \times 10^{23} \text{ atoms}}{1 \text{ mole}} = 6.02 \times 10^{24} \text{ atoms Ti}$$

All Mixed Up Chemistry Worksheet Answers

Ian Pickup



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About the Author: DR DENNIS ENGLIN earned his bachelor's from Westmont College, his master of science from California State University, and his EdD from the University of Southern California. He enjoys teaching animal biology, vertebrate biology, wildlife biology, organismic biology, and astronomy at The Master's University. His professional memberships include the Creation Research Society, the American Fisheries Association, Southern California Academy of Sciences, Yellowstone Association, and Au Sable Institute of Environmental Studies.

Addison-Wesley Science Insights, 1996

Exploring Earth and Space Michael DiSpezio, 1995 A textbook exploring such aspects of matter and energy as heat, electricity, and nuclear chemistry with suggested activities and review questions at the end of each chapter.

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web in the beginning before all time was the word christ and the word was with god and the word was god himself christian standard bible in the beginning was the word and the word was with god and the word was god

word choice at the beginning or in the beginning english - Sep 05 2023

web oct 18 2012 i think the most important difference is that in the beginning seems to be an expression describing a whole period of time while at the beginning more literally describes a single moment in time similar to the difference between saying

genesis 1 kjv in the beginning god created the heaven bible gateway - Jun 02 2023

web 1 in the beginning god created the heaven and the earth 2 and the earth was without form and void and darkness was upon the face of the deep and the spirit of god moved upon the face of the waters 3 and god said let there be light and there was light 4 and god saw the light that it was good and god divided the light from the darkness

genesis 1 niv the beginning in the beginning god bible gateway - Oct 06 2023

web the beginning 1 in the beginning a god created b the heavens c and the earth d 2 now the earth was formless e and empty f darkness was over the surface of the deep g and the spirit of god h was hovering i over the waters 3 and god said j let there be light and there was light

in the beginning phrase wikipedia - Jul 03 2023

web in the beginning phrase the first chapter of b reshith or genesis written on an egg in the jerusalem museum in the beginning of bereshith in biblical hebrew is the opening phrase or incipit used in the bible in genesis 1 1 in john 1 1 of the new testament the word archē is translated into english with the same phrase etymology