



Analytical Chemistry Interview Questions And Answers Guide

- Q1:** What is the difference between qualitative and quantitative analysis?
- A1:** Qualitative analysis is the determination of the presence or absence of a particular element or compound in a sample. Quantitative analysis is the determination of the amount or concentration of a particular element or compound in a sample.
- Q2:** What is the difference between a standard solution and a sample solution?
- A2:** A standard solution is a solution of known concentration of a particular substance. A sample solution is a solution of unknown concentration of a particular substance.
- Q3:** What is the difference between a primary standard and a secondary standard?
- A3:** A primary standard is a substance that is pure, stable, and has a high molecular weight. A secondary standard is a substance that is not pure, stable, or has a low molecular weight.
- Q4:** What is the difference between a titrant and an analyte?
- A4:** A titrant is a solution of known concentration that is used to determine the concentration of an analyte. An analyte is a substance whose concentration is to be determined.
- Q5:** What is the difference between a standard deviation and a relative standard deviation?
- A5:** A standard deviation is a measure of the spread of data. A relative standard deviation is a measure of the spread of data relative to the mean.
- Q6:** What is the difference between a calibration curve and a standard curve?
- A6:** A calibration curve is a plot of the response of an analytical instrument versus the concentration of a standard solution. A standard curve is a plot of the response of an analytical instrument versus the concentration of a sample solution.
- Q7:** What is the difference between a blank and a sample?
- A7:** A blank is a sample that contains no analyte. A sample is a solution of unknown concentration of a particular substance.
- Q8:** What is the difference between a reagent and a standard?
- A8:** A reagent is a substance that is used in a chemical reaction. A standard is a substance of known concentration that is used to determine the concentration of an analyte.
- Q9:** What is the difference between a titration and a standardization?
- A9:** A titration is a process of determining the concentration of an analyte by reacting it with a known amount of a titrant. A standardization is a process of determining the concentration of a standard solution by reacting it with a known amount of an analyte.
- Q10:** What is the difference between a primary standard and a secondary standard?
- A10:** A primary standard is a substance that is pure, stable, and has a high molecular weight. A secondary standard is a substance that is not pure, stable, or has a low molecular weight.

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