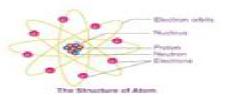
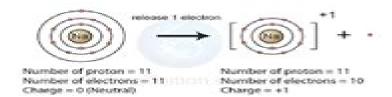
Chapter 2: The Chemistry of Life

- Legos of the Universe
 - o Atoms
 - Elements
 - Gold, carbon, silver, etc. (periodic table)
- > Atomic Structure
 - Nucleus
 - Proton (+)
 - Neutrons (no charge)
 - o Electron Shell
- > Isotopes
 - Atomic Mass
- Radioactive Atoms
 - Some atomic nuclei are not stable and break down randomly
 - These atoms are radioactive
 - During decomposition they release, at a constant rate, a tiny high-speed particle carrying a lot of energy
- > Electron shells determine bonding patterns
- Versatility of Carbon
 - Can make 4 bonds
 - Most often with C,H,O, and N
 - Huge potential variety
- > Ions
 - Particles in the outer shell
- > Changing Parts of the Atom
 - What happens if you...
 - Add an electron to chlorine
 - Add a neutron to nitrogen
 - Add a proton to hydrogen
- > Elements in Living Things
 - Oxygen
 - Carbon
 - Hydrogen
 - Nitrogen
- ➤ Molecules
 - Molecules
 - Bond energy
 - Chemical reactions
 - Reactants > Products
- Chemical Bonding
 - Covalent
 - o Ionie







Answer Key To The Chemistry Of Life

Kevin T. Patton, Gary A.
Thibodeau, Linda Swisher

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