

Molecular Polarity Lab

Name _____ Per. _____

Circle the best correct choice. Check the other correct choice(s) if indicated.

Part A: Three atom investigations

1. Cl₂ (Cl-Cl) has the best fit bond dipole moment and bond strengths for about three-dimensional and the bond length value
 - a. 0.00 D and 243 pm
 - b. 0.00 D and 199 pm
 - c. 0.00 D and 127 pm
 - d. 0.00 D and 99 pm
2. Draw the bond dipole moment of Cl₂ and indicate the electronegativity of each Cl.

Observe the arrow, partial charges and bond structure.

Write the following information: the bond dipole moment of Cl₂ is about 0.00 D.

at the _____

partial charges _____

bond structure _____
3. Draw partial bond dipole moment for a polar molecule containing the electronegativity of B and observe the results. Fill in the following information:
 - a. The polarity arrow points to the _____ electronegativity atom.
 - b. The partial positive charge is always on the _____ electronegativity atom.
 - c. The larger the electronegativity difference most _____ the bond structure.

Part B: Three atom investigations

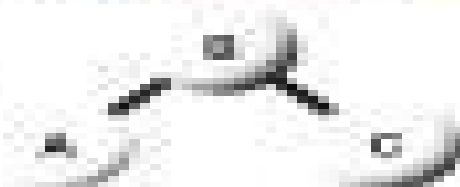
1. Draw the correct bond dipole moment for atoms and the electronegativity of Cl and C and Cl and H.

Observe the bond dipole moment, molecular dipole and partial charges.

 - a. Draw the bond dipole on the Lewis dot diagram.
 - b. In another set up, observe partial molecular dipole on the diagram.
 - c. Draw the partial charge systematic position on the diagram.
 - d. Circle the electronegativity. Add other electronegativity the molecule is represented.
2. Circle the best choice.  Draw the bond dipole moment for atoms and the electronegativity of C and H and Cl and H.

Observe the bond dipole moment, molecular dipole and partial charges.

 - a. Draw the bond dipole on the Lewis dot diagram.
 - b. In another set up, observe partial molecular dipole on the diagram.
 - c. Draw the partial charge systematic position on the diagram.
 - d. Circle the electronegativity. Add other electronegativity the molecule is represented.



Answer Of Molecular Polarity Phet Lab

Marco Cascella



Answer Of Molecular Polarity Phet Lab:

Simulations and Student Learning Matthew Schnurr, Anna MacLeod, 2021-01-04 The book underlines the value of simulation based education as an approach that fosters authentic engagement and deep learning **Learning and Performance Assessment: Concepts, Methodologies, Tools, and Applications** Management Association, Information Resources, 2019-10-11 As teaching strategies continue to change and evolve and technology use in classrooms continues to increase it is imperative that their impact on student learning is monitored and assessed New practices are being developed to enhance students participation especially in their own assessment be it through peer review reflective assessment the introduction of new technologies or other novel solutions Educators must remain up to date on the latest methods of evaluation and performance measurement techniques to ensure that their students excel Learning and Performance Assessment Concepts Methodologies Tools and Applications is a vital reference source that examines emerging perspectives on the theoretical and practical aspects of learning and performance based assessment techniques and applications within educational settings Highlighting a range of topics such as learning outcomes assessment design and peer assessment this multi volume book is ideally designed for educators administrative officials principals deans instructional designers school boards academicians researchers and education students seeking coverage on an educator s role in evaluation design and analyses of evaluation methods and outcomes **Handbook of Research on Innovative Pedagogies and Technologies**

for Online Learning in Higher Education Vu, Phu, Fredrickson, Scott, Moore, Carl, 2016-12-28 The integration of technology has become an integral part of the educational environment By developing new methods of online learning students can be further aided in reaching goals and effectively solving problems The Handbook of Research on Innovative Pedagogies and Technologies for Online Learning in Higher Education is an authoritative reference source for the latest scholarly research on the implementation of instructional strategies tools and innovations in online learning environments Featuring extensive coverage across a range of relevant perspectives and topics such as social constructivism collaborative learning and projects and virtual worlds this publication is ideally designed for academicians practitioners and researchers seeking current research on best methods to effectively incorporate technology into the learning environment

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