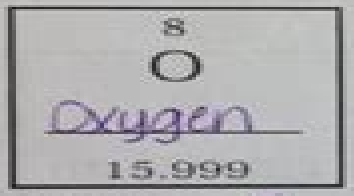
The Atoms Family
Atomic Math Challenge

Name _____

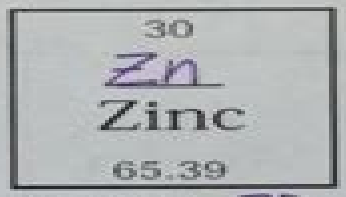
Atomic Number
Symbol
Name
Atomic Mass

Atomic number equals the number of protons or electrons

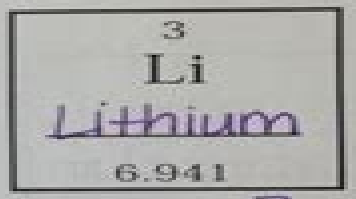
Atomic mass equals the number of protons + neutrons



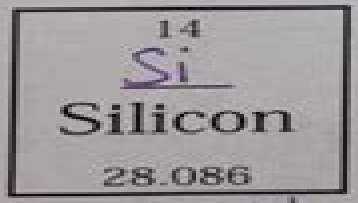
Atomic # = 8
Atomic Mass = 16
# of Protons = 8
# of Neutrons = 8
# of Electrons = 8



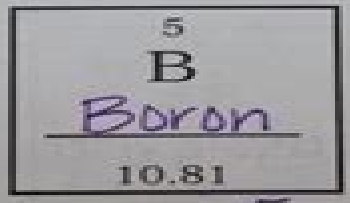
Atomic # = 30
Atomic Mass = 65
# of Protons = 30
# of Neutrons = 35
# of Electrons = 30



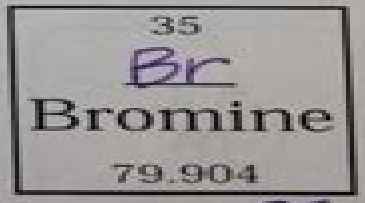
Atomic # = 3
Atomic Mass = 7
# of Protons = 3
# of Neutrons = 4
# of Electrons = 3



Atomic # = 14
Atomic Mass = 28
# of Protons = 14
# of Neutrons = 14
# of Electrons = 14



Atomic # = 5
Atomic Mass = 11
# of Protons = 5
# of Neutrons = 6
# of Electrons = 5



Atomic # = 35
Atomic Mass = 80
# of Protons = 35
# of Neutrons = 45
# of Electrons = 35

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M Walker

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