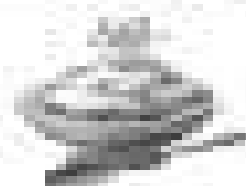


Atomic Probing Structure of the Atom

Name _____
Date _____



Purpose:

To determine which elements are present in a sample of an element by using the emission spectrum.

Part I: Finding Wavelengths in Emission Spectra

When an element is excited, it emits light. The light is composed of many different wavelengths. The light is called a continuous spectrum. The light is called a continuous spectrum because the wavelengths are continuous. The light is called a continuous spectrum because the wavelengths are continuous.

Procedure: In the laboratory, a sample of an element is excited. The light is called a continuous spectrum. The light is called a continuous spectrum because the wavelengths are continuous.

The light is called a continuous spectrum because the wavelengths are continuous. The light is called a continuous spectrum because the wavelengths are continuous.

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Sabine Zange



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