

Ternary compounds have _____ (three) _____ (three or) elements. They are usually always composed of a metal(s) _____ (metals) anionic ion(s) and a polyatomic _____ (polyatomic) anion. The only positively charged polyatomic ion is _____ (ammonium). To write the formula of a ternary compound is no different than to write the formula of a binary compound with one exception. If a subscript is necessary for the polyatomic ion in order to _____ (balance) charge, we must place the polyatomic ion in _____ (parentheses).

As each ion, write the formula of the ionic compound consisting of the cation on the left of the formula and the negative ion(s) on the right.

	OH^-	SO_4^{2-}	NO_3^-	CO_3^{2-}	PO_4^{3-}
K^+	_____	_____	_____	_____	_____
Na^{2+}	_____	_____	_____	_____	_____
NH_4^+	_____	_____	_____	_____	_____
Al^{3+}	_____	_____	_____	_____	_____
Ca^{2+}	_____	_____	_____	_____	_____
Fe^{2+}	_____	_____	_____	_____	_____

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