#### Bachelor of Engineering Examination, 2023

#### (1st Year, 1st Semester)

#### CHEMISTRY

Time: 3 hours

Full Marks: 100

Use separate answer sheet for Chemistry-I, Chemistry-II and Chemistry-III

# CO-ICHEMISTRY-I Answer any three questions from CO-1 O1. (a) Derive the expression for the calculation of the radius of Bohr's orbit and the velocity of the electron in hydrogen atom. 2+2 (b) What would be the wavelength of the 2<sup>nd</sup> line in the Lyman series for the H atom [Given $R_{SI} = 109678 \text{ cm}^{-1}$ ]. (c) Calculate the de Broglie wavelength of a bullet having mass = $2 \times 10^{-3}$ kg, moving with the speed of 300 ms<sup>-1</sup>. (d) State and explain "Heisenberg uncertainty principle". 2 O2. (a) What are the merits and demerits of Bohr's atomic model? 3 (b) Calculate the ionization energy of Li<sup>2+</sup> ion. T. 5 1 + 2.5(c) Define lattice energy. Construct Born-Haber cycle of NaCl. (d) Applying Fajan's rule explain the trend on thermal stabilities of the following compounds . BeCO3, MgCO3, CaCO3, SrCO3, BaCO3. 2 (3. (a) Define buffer solutions with examples. (b) Discuss the mechanism of buffer action. (c) Determine the pH of a solution obtained by mixing equal volumes of 0.1 M NH4Cl and 0.02 M NH4OH. [Given that Kb for NH4OH is 1.8 x 10-5]

Q4. (a) Explain the trend of H-X-H angles (X = central atom) in CH4, NH3 and H2O.

# **Answer To 1st Semester Chemistry Exam Review**

Richard H. Langley, John Moore

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