

Name _____ Date _____ Class _____

CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

Matter—Properties and Changes

Section 3.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical density	mass properties	physical substance
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Matter is anything with (1) mass and volume. A (2) substance is a form of matter with a uniform and unchanging composition. Substances have specific, unchanging (3) properties that can be observed. Substances have both physical and chemical properties. (4) Physical properties can be observed without changing a substance's chemical composition. Color, hardness, and (5) density are examples. Other properties cannot be observed without changing the composition of a substance. These are called (6) chemical properties. An example is the tendency of iron to form rust when exposed to air.

Label each property as either physical or chemical.

- physical 7. Chemical formula H_2O
- chemical 8. Irons react vigorously when exposed to moist air
- chemical 9. Remains unchanged when in the presence of nitrogen
- physical 10. Colorless
- physical 11. Solid at normal temperatures and pressures
- chemical 12. Ability to combine with another substance
- physical 13. Melting point
- physical 14. Liquid at normal temperatures and pressures
- physical 15. Boiling point is 100°C
- physical 16. Conducts electricity
- physical 17. Density is $\frac{1}{2} \frac{\text{g}}{\text{cm}^3}$

Name _____ Date _____ Class _____

CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

Section 3.1 continued

In your textbook, read about states of matter.

Label each drawing with one of these words: solid, liquid, gas.



For each statement below, write true or false.

- true 21. All matter that we encounter in everyday life exists as one of these physical forms.
- true 22. A solid has definite shape and volume.
- false 23. A liquid has a definite shape and takes on the volume of its container.
- true 24. A gas has both the shape and the volume of its container.
- false 25. The particles in a gas cannot be compressed into a smaller volume.
- false 26. Liquids tend to contract when heated.
- false 27. The particles in a solid are spaced far apart.
- false 28. The words gas and vapor can be used interchangeably.

Answers Chemistry Matter And Change

David Kirk



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Table of Contents Answers Chemistry Matter And Change

1. Understanding the eBook Answers Chemistry Matter And Change
 - The Rise of Digital Reading Answers Chemistry Matter And Change
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers Chemistry Matter And Change
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers Chemistry Matter And Change
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers Chemistry Matter And Change
 - Personalized Recommendations
 - Answers Chemistry Matter And Change User Reviews and Ratings

- Answers Chemistry Matter And Change and Bestseller Lists
- 5. Accessing Answers Chemistry Matter And Change Free and Paid eBooks
 - Answers Chemistry Matter And Change Public Domain eBooks
 - Answers Chemistry Matter And Change eBook Subscription Services
 - Answers Chemistry Matter And Change Budget-Friendly Options
- 6. Navigating Answers Chemistry Matter And Change eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers Chemistry Matter And Change Compatibility with Devices
 - Answers Chemistry Matter And Change Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers Chemistry Matter And Change
 - Highlighting and Note-Taking Answers Chemistry Matter And Change
 - Interactive Elements Answers Chemistry Matter And Change
- 8. Staying Engaged with Answers Chemistry Matter And Change
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers Chemistry Matter And Change
- 9. Balancing eBooks and Physical Books Answers Chemistry Matter And Change
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers Chemistry Matter And Change
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers Chemistry Matter And Change
 - Setting Reading Goals Answers Chemistry Matter And Change
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers Chemistry Matter And Change
 - Fact-Checking eBook Content of Answers Chemistry Matter And Change
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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