

1 Complete the word equations for the reactions of the Group 7 elements with metals.

- a iron + chlorine → iron chloride
 b iron + bromine → iron bromide
 c sodium + iodine → sodium iodide
 d potassium + bromine → potassium bromide
 e lithium + chlorine → lithium chloride
 f sodium + chlorine → sodium chloride

2 Make up four sentences using the phrases in the table. Each sentence must include a phrase from each column.

There are	weak forces	between the two atoms of a chlorine molecule.
	strong forces	between a chlorine molecule and its neighbours.
		between the two atoms of an iodine molecule.
		between an iodine molecule and its neighbours.

*There are weak forces between the two atoms of a chlorine molecule.
 There are strong forces between a chlorine molecule and its neighbours.
 There are weak forces between the two atoms of an iodine molecule.
 There are strong forces between an iodine molecule and its neighbours.*

3 For the sentences A–H below:

- Write **1** next to the sentences that are true for Group 1.
 - Write **7** next to the sentences that are true for Group 7.
 - Write **B** next to the sentences that are true for both Groups 1 and 7.
- A Going down this group, boiling point increases.
 B Going up this group, the reactions of the elements get more vigorous.
 C Going down this group, the number of protons in atoms of the elements increases.
 D Atoms of the elements in this group form molecules made up of two atoms.
 E Going up this group, the colours of the elements get lighter.
 F The elements at the top of this group are gases.
 G The elements in this group conduct electricity.
 H All the atoms of the elements in this group have the same number of electrons in the outermost shell.

7
1
B
1
B
7
B

12 Write **M** next to the pair of elements that you predict will react most vigorously.
 Write **L** next to the pair of elements that you predict will react least vigorously.

- a sodium and iodine
 b lithium and iodine
 c potassium and fluorine
 d potassium and bromine
 e sodium and bromine

L
L
M
M
L

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Table of Contents Answers For A Ls Chemistry

1. Understanding the eBook Answers For A Ls Chemistry
 - The Rise of Digital Reading Answers For A Ls Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For A Ls Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For A Ls Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For A Ls Chemistry
 - Personalized Recommendations
 - Answers For A Ls Chemistry User Reviews and Ratings
 - Answers For A Ls Chemistry and Bestseller Lists
5. Accessing Answers For A Ls Chemistry Free and Paid eBooks

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 - Answers For A Ls Chemistry Budget-Friendly Options
6. Navigating Answers For A Ls Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers For A Ls Chemistry Compatibility with Devices
 - Answers For A Ls Chemistry Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For A Ls Chemistry
 - Highlighting and Note-Taking Answers For A Ls Chemistry
 - Interactive Elements Answers For A Ls Chemistry
 8. Staying Engaged with Answers For A Ls Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For A Ls Chemistry
 9. Balancing eBooks and Physical Books Answers For A Ls Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For A Ls Chemistry
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Answers For A Ls Chemistry
 - Setting Reading Goals Answers For A Ls Chemistry
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Answers For A Ls Chemistry
 - Fact-Checking eBook Content of Answers For A Ls Chemistry
 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

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- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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