

Experiment 17 Report Sheet

Antacid Analysis

Date _____ Lab Sec. _____ Name _____ Desk No. _____

A. Dissolving the Antacid

	Trial 1	Trial 2	Trial 1	Trial 2
1. Mass of flask (g)	128.457	97.342	115.395	111.220
2. Mass of flask + antacid sample (g)	128.661	97.542	115.606	111.440
3. Mass (or tared mass) of antacid sample (g)	.204	.200	0.211	0.217
4. Total volume of HCl added (mL)	25	25	25	25
5. Molar concentration of HCl (mol/L)	.09687		0.9687	

B. Analyzing the Antacid Sample

1. Molar concentration of NaOH (mol/L)	.09361		0.9361	
2. Buret reading, initial (mL)	1.5	16.7	9.7	11.6
3. Buret reading, final (mL)	16.7	28.4	18.2	21.7
4. Volume of NaOH (mL)	15.2	12.3	8.5	10.1

C. Calculations

1. Moles of HCl added, total (mol)				
2. Moles of NaOH added (mol)				
3. Moles of base in antacid sample (mol)				
4. $\frac{\text{mol base in antacid}}{\text{mass of antacid sample}}$ (mol/g)		*		*
5. Average $\frac{\text{mol base in antacid}}{\text{mass of antacid sample}}$ (mol/g)				

*Show calculation(s) for Trial(s) 1 on the next page.

Analysis Of Antacids Lab Report Answers

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Staley, Michael S. Matta, Edward L.
Waterman, Prentice-Hall Staff**



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