## Analytical Chemistry MCQs: TYBSC SEM III

 The solubility product K<sub>sp</sub> of sparingly soluble Ag<sub>2</sub>CrO<sub>4</sub> is 4\*10<sup>-12</sup>. The molar solubility of the salt is

- (a) 1.0\*10\* moles/lit
- (b) 2.0\*104 moles/lit
- (c) 1.0\*10-5 moles/lit
- (d) 1.2\*10-12 moles/lit

 Solubility product of AgCl is 1.44\*10<sup>-4</sup> moles/lit at 373K. The solubility of the AgCl is boiling of water will be....

- (a) 0.72\*104 moles/lit
- (b) 0.72\*10-2 moles/lit
- (c) 1.20\*10-2 moles/lit
- (d)1.20\*10<sup>-4</sup> moles/lit
- 10. Precipitation takes place when the ionic product...
- (a) Is zero
- (b) Exceeds the solubilty product
- (c) Equal to solubilty product
- (d) Is less than the solubility product
- 11. The electomotive force is expressed In the unit of .....
- (a) Ohm
- (b) Volt
- (c) Ampere
- (d) Colulomb
- 12. If the cell can provide electrical energy to an external system, it is called as
- (a) Galvanic cell
- (b) Electrolytic cell
- (c) Concentration cell
- (d) Chemical cell
- The current per unit area of electrode surface is known as....
- (a) Current efficiency
- (b) Current yield
- (e) Current density
- (d) None of these

# **Analytical Chemistry Mcqs**

**Arshad Iqbal** 

#### **Analytical Chemistry Mcqs:**

MCQs Chemistry V. K. Ahluwalia, 2013-12-30 This book has been divided in 37 chapters under three parts Inorganic Chemistry Organic Chemistry and Physical Chemistry for convenient understanding It also includes solved model test papers of the previous three years of AIIMS CBSE PMT CPMT UP to enable students to develop the skills of problem solving and time management essential for any entrance examination In addition to providing answers to all the questions detailed explanatory notes to selected difficult questions have also been provided to justify the answer A separate section of Assertions and Reasons is also given at the end of each chapter Exhaustive Question Bank Explanatory Notes and Hints Assertions Reasons Includes Pre solved papers of five years Models Test Papers of AIIMS CBSE PMT CPMT Chemistry MCQ (Multiple Choice Questions) Arshad Igbal, 2019-06-18 The A Level Chemistry Multiple Choice Questions MCQ Quiz with Answers PDF A Level Chemistry MCQ PDF Download Quiz Questions Chapter 1 28 Practice Tests with Answer Key IGCSE GCE Chemistry Questions Bank MCQs Notes includes revision guide for problem solving with hundreds of solved MCQs A Level Chemistry MCQ with Answers PDF book covers basic concepts analytical and practical assessment tests A Level Chemistry MCQ PDF book helps to practice test questions from exam prep notes The A Level Chemistry MCQs with Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved MCQs A Level Chemistry Multiple Choice Questions and Answers MCQs PDF Free download chapter 1 a book covers solved guiz questions and answers on chapters Alcohols and esters atomic structure and theory benzene chemical compound carbonyl compounds carboxylic acids acyl compounds chemical bonding chemistry of life electrode potential electrons in atoms enthalpy change equilibrium group IV groups II and VII halogenoalkanes hydrocarbons introduction to organic chemistry ionic equilibria lattice energy moles and equations nitrogen and sulfur organic and nitrogen compounds periodicity polymerization rates of reaction reaction kinetics redox reactions and electrolysis states of matter transition elements tests for college and university revision guide A Level Chemistry Quiz Questions and Answers PDF free download eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The book IGCSE GCE Chemistry MCQs Chapter 1 28 PDF includes high school question papers to review practice tests for exams A Level Chemistry Multiple Choice Questions MCQ with Answers PDF digital edition eBook a study guide with textbook chapters tests for IGCSE NEET MCAT GRE GMAT SAT ACT competitive exam A Level Chemistry Mock Tests Chapter 1 28 eBook covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Alcohols and Esters MCQ Chapter 2 Atomic Structure and Theory MCQ Chapter 3 Benzene Chemical Compound MCQ Chapter 4 Carbonyl Compounds MCQ Chapter 5 Carboxylic Acids and Acyl Compounds MCQ Chapter 6 Chemical Bonding MCQ Chapter 7 Chemistry of Life MCQ Chapter 8 Electrode Potential MCQ Chapter 9 Electrons in Atoms MCQ Chapter 10 Enthalpy Change MCQ Chapter 11 Equilibrium MCQ Chapter 12 Group IV MCQ Chapter 13 Groups II and VII MCQ Chapter 14 Halogenoalkanes MCQ Chapter 15 Hydrocarbons MCQ Chapter 16

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