

CHAPTER 7 REVIEW

Chemical Formulas and Chemical Compounds**SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. _____ In a Stock system name such as iron(III) sulfate, the Roman numeral tells us
 - (a) how many atoms of Fe are in one formula unit.
 - (b) how many sulfate ions can be attached to the iron atom.
 - (c) the charge on each Fe ion.
 - (d) the total positive charge of the formula unit.
2. _____ Changing a subscript in a correctly written chemical formula
 - (a) changes the number of moles represented by the formula.
 - (b) changes the charges on the other ions in the compound.
 - (c) changes the formula so that it no longer represents the compound it previously represented.
 - (d) has no effect on the formula.
3. The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.
 - _____ a. How many elements make up this compound?
 - _____ b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$?
 - _____ c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$?
 - _____ d. How many atoms are present in a sample of 2.0×10^{23} molecules of $C_7H_5(NO_2)_3$?
4. How many atoms are present in each of these formula units?
 - _____ a. $Ca(HCO_3)_2$
 - _____ b. $C_{12}H_{22}O_{11}$
 - _____ c. $Fe(ClO_2)_3$
 - _____ d. $Fe(ClO_3)_2$
5. _____
 - _____ a. What is the formula for the compound dinitrogen pentoxide?
 - _____ b. What is the Stock system name for the compound FeO ?
 - _____ c. What is the formula for sulfurous acid?
 - _____ d. What is the name for the acid H_3PO_4 ?

Answers Review Chemical Formulas And Chemical Compounds

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Compounds Questions Chapter 5 Carboxylic Acids and Acyl Compounds Questions Chapter 6 Chemical Bonding Questions
Chapter 7 Chemistry of Life Questions Chapter 8 Electrode Potential Questions Chapter 9 Electrons in Atoms Questions
Chapter 10 Enthalpy Change Questions Chapter 11 Equilibrium Questions Chapter 12 Group IV Questions Chapter 13
Groups II and VII Questions Chapter 14 Halogenoalkanes Questions Chapter 15 Hydrocarbons Questions Chapter 16
Introduction to Organic Chemistry Questions Chapter 17 Ionic Equilibria Questions Chapter 18 Lattice Energy Questions
Chapter 19 Moles and Equations Questions Chapter 20 Nitrogen and Sulfur Questions Chapter 21 Organic and Nitrogen
Compounds Questions Chapter 22 Periodicity Questions Chapter 23 Polymerization Questions Chapter 24 Rates of Reaction
Questions Chapter 25 Reaction Kinetics Questions Chapter 26 Redox Reactions and Electrolysis Questions Chapter 27 States
of Matter Questions Chapter 28 Transition Elements Questions The Alcohols and Esters Quiz Questions PDF e Book Chapter
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Table of Contents Answers Review Chemical Formulas And Chemical Compounds

1. Understanding the eBook Answers Review Chemical Formulas And Chemical Compounds
 - The Rise of Digital Reading Answers Review Chemical Formulas And Chemical Compounds
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers Review Chemical Formulas And Chemical Compounds
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers Review Chemical Formulas And Chemical Compounds
 - User-Friendly Interface

4. Exploring eBook Recommendations from Answers Review Chemical Formulas And Chemical Compounds
 - Personalized Recommendations
 - Answers Review Chemical Formulas And Chemical Compounds User Reviews and Ratings
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5. Accessing Answers Review Chemical Formulas And Chemical Compounds Free and Paid eBooks
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8. Staying Engaged with Answers Review Chemical Formulas And Chemical Compounds
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 - Minimizing Distractions
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 - Setting Reading Goals Answers Review Chemical Formulas And Chemical Compounds
 - Carving Out Dedicated Reading Time

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