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I Am Error Nathan Altice, 2017-09-08 The complex material histories of the Nintendo Entertainment System platform from code to silicon focusing on its technical constraints and its expressive affordances In the 1987 Nintendo Entertainment System videogame *Zelda II The Adventure of Link* a character famously declared I AM ERROR Puzzled players assumed that this cryptic message was a programming flaw but it was actually a clumsy Japanese English translation of My Name is Error a benign programmer's joke In *I AM ERROR* Nathan Altice explores the complex material histories of the Nintendo Entertainment System and its Japanese predecessor the Family Computer offering a detailed analysis of its programming and engineering its expressive affordances and its cultural significance Nintendo games were rife with mistranslated texts but as Altice explains Nintendo's translation challenges were not just linguistic but also material with consequences beyond simple misinterpretation Emphasizing the technical and material evolution of Nintendo's first cartridge based platform Altice describes the development of the Family Computer or Famicom and its computational architecture the translation problems faced while adapting the Famicom for the U S videogame market as the redesigned Entertainment System Nintendo's breakthrough console title *Super Mario Bros* and its remarkable software innovations the introduction of Nintendo's short lived proprietary disk format and the design repercussions on *The Legend of Zelda* Nintendo's efforts to extend their console's lifespan through cartridge augmentations the Famicom's Audio Processing Unit APU and its importance for the chiptunes genre and the emergence of software emulators and the new kinds of play they enabled

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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable measure an appropriate volume of each liquid eg 25 cm³ 3 place one of the

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with HCl as 57.9 kJ mol⁻¹

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web the enthalpy change of neutralisation is defined as the enthalpy change when solutions of an acid and alkali react together to produce one mole of water for example HCl(aq) + NaOH(aq) → NaCl(aq) + H₂O(aq) you can see that one mole of water is produced when one mole of HCl or one mole of NaOH reacts

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