



General Chemistry II Equation Sheet

Think about how to set up the problem first, then apply the needed principles and formulas.

Phase Changes

$$\Delta T_b = iK_b m$$

$$\Delta T_f = iK_f m$$

$$\pi = iMRT$$

$$c = kP$$

$$\frac{P_1}{P_2} = \frac{Z_1}{Z_2}$$

$$P_{\text{solution}} = \chi_{\text{solvent}} P_{\text{solvent}}^{\circ}$$

$$\Delta P = P_{\text{solvent}}^{\circ} - P_{\text{solution}} = \chi_{\text{solute}} P_{\text{solvent}}^{\circ}$$

Cells

$$E = E^{\circ} - \frac{RT}{nF} \ln(Q)$$

$$E = E^{\circ} - \frac{0.0592}{n} \log(Q) \quad (\text{at } 25^{\circ}\text{C})$$

$$\Delta G = -nFE_{\text{cell}}$$

$$\Delta G^{\circ} = -nFE_{\text{cell}}^{\circ}$$

$$E_{\text{cell}}^{\circ} = \frac{RT}{nF} \ln(K)$$

$$E_{\text{cell}}^{\circ} = \frac{0.0592\text{V}}{n} \log(K) \quad (\text{at } 25^{\circ}\text{C})$$

USE ONE:

$$E_{\text{cell}}^{\circ} = E_{\text{cathode(re.d.)}}^{\circ} - E_{\text{anode(re.d.)}}^{\circ}$$

$$E_{\text{cell}}^{\circ} = E_{\text{cathode(re.d.)}}^{\circ} + E_{\text{anode(ox.)}}^{\circ}$$

Thermodynamics

$$\Delta G = \Delta H - T\Delta S$$

$$\Delta S_{\text{surr}} = \frac{-\Delta H_{\text{sys}}}{T}$$

$$\Delta S_{\text{univ}} = \Delta S_{\text{sys}} + \Delta S_{\text{surr}}$$

$$\Delta S_{\text{ran}}^{\circ} = \Sigma S_{\text{products}}^{\circ} - \Sigma S_{\text{reactants}}^{\circ}$$

$$\Delta G_{\text{ran}}^{\circ} = \Sigma \Delta G_{\text{products}}^{\circ} - \Sigma \Delta G_{\text{reactants}}^{\circ}$$

Concentrations

$$\text{molality} = \frac{\text{moles solute}}{\text{mass solvent (kg)}}$$

$$\% \text{ by mass} = \frac{\text{moles solute}}{\text{moles solute} + \text{moles solvent}} \times 100\%$$

$$\text{parts per million} = \frac{\text{moles solute}}{\text{moles solute} + \text{moles solvent}} \times 10^6$$

$$\text{parts per billion} = \frac{\text{moles solute}}{\text{moles solute} + \text{moles solvent}} \times 10^9$$

Acids and Bases

$$K_w = [H_3O^+][OH^-] = K_a \times K_b$$

$$pOH = -\log[OH^-]$$

$$[OH^-] = 10^{-pOH}$$

$$pH + pOH = pK_a + pK_b = 14.00 \quad (\text{at } 25^{\circ}\text{C})$$

$$pH = pK_a + \log \frac{[A^-]}{[HA]}$$

$$pOH = pK_b + \log \frac{[BH^+]}{[B]}$$

$$\% \text{ ionization} = \frac{[H_3O^+]_{\text{eq}}}{[HA]_{\text{eq}}} \times 100\%$$

$$pK_a = -\log(K_a)$$

$$pK_b = -\log(K_b)$$

$$K_a = 10^{-pK_a}$$

$$K_b = 10^{-pK_b}$$

Equilibria

$$Q_c = \frac{[C]^c[D]^d}{[A]^a[B]^b}$$

$$K_c = \frac{[C]^c[D]^d}{[A]^a[B]^b}$$

$$Q_p = \frac{P_C^c P_D^d}{P_A^a P_B^b}$$

$$K_p = \frac{P_C^c P_D^d}{P_A^a P_B^b}$$

$$K_p = K_c [R \times T]^{\Delta n}$$

$$\Delta G = \Delta G^{\circ} + RT \ln(Q)$$

$$\Delta G^{\circ} = -RT \ln(K)$$

$$\text{rate} = k[A]^x[B]^y$$

$$k = Ae^{-\frac{E_a}{RT}}$$

$$\ln\left(\frac{k_2}{k_1}\right) = \frac{E_a}{R} \left(\frac{1}{T_1} - \frac{1}{T_2}\right)$$

$$\text{half life} = \frac{\ln(2)}{k} \quad (\text{1st order reactions only})$$

$$[A]_t = -kt + [A]_0$$

$$\ln([A]_t) = -kt + \ln([A]_0)$$

$$\frac{1}{[A]_t} = kt + \frac{1}{[A]_0}$$

Answer Sheet General Chemistry Equations

R Barnett



Answer Sheet General Chemistry Equations:

General Chemistry Workbook Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material Chemical Symbolism and Calculations Stanley Wallace Morse, 1953 **Laboratory Manual**

for Principles of General Chemistry Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

Laboratory Manual for Principles of General Chemistry J. A. Beran, Mark Lassiter, 2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don't write experiments A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered Further experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition's experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each experiment's Report Sheet to provide a space for recording observations and data during the experiment Continued emphasis on handling data is supported by the Data Analysis section General Chemistry Donald A. McQuarrie, Stanley Gill, 2011-06-15 This Fourth Edition of McQuarrie's classic text offers a thorough revision and a quantum leap forward from the previous edition Taking an atoms first approach it promises to be another ground breaking text in the tradition of McQuarrie's many previous works This outstanding new text available in a soft cover edition offers professors a fresh choice and outstanding value Regents Exams and Answers: Chemistry--Physical Setting Revised Edition Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students

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Basic Concepts of Chemistry Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features Each section focuses on the development and assessment of one or two specific objectives Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives These features are followed

by a set of independent practice problems The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance *Princeton Review MCAT General Chemistry Review, 5th Edition* The Princeton Review, 2025-11-04 Everything students need to know to succeed on the General Chemistry section of the MCAT exam Medical College Admission Test including 3 full length practice tests With in depth coverage of all MCAT general chemistry topics The Princeton Review MCAT General Chemistry Review helps you practice your way to excellence featuring everything you need for a high score Thorough content review for of all MCAT general chemistry topics 3 full length online practice tests with detailed answer explanations and score reports Test taking strategies geared toward general chemistry mastery Bulleted chapter summaries for quick review Full color illustrations diagrams and tables End of chapter drills and explanations plus MCAT style passages and questions 3 scholarly journal articles with exercises to strengthen your critical analysis Extensive reference glossary The Princeton Review MCAT General Chemistry Review covers Chemistry Fundamentals Atomic Structure and Periodic Trends Bonding and Intermolecular Forces Thermodynamics Phases Gases Kinetics Equilibrium Acids and Bases Eletrochemistry MCAT Math for General Chemistry Basic Concept of Inorganic Chemistry Mr. Ajeet Singh, 2018-05-22 EduGorilla has its own publishing wing producing exam prep books trade books etc *Admission Assessment Exam Review - E-Book* HESI, 2024-03-26 Taking the HESI Admission Assessment A2 Exam is the first step on the journey to becoming a successful healthcare professional Be prepared to pass the exam with the most up to date HESI Admission Assessment Exam Review 6th Edition From the testing experts at HESI this user friendly guide walks you through the topics and question types found on admission exams including math reading comprehension vocabulary grammar biology chemistry and anatomy and physiology The guide includes hundreds of sample questions as well as step by step explanations illustrations and comprehensive practice exams to help you review various subject areas and improve test taking skills Plus the pre test and post test help identify your specific weak areas so study time can be focused where it s needed most NEW Updated thoroughly reviewed content helps you prepare for the HESI Admission Assessment Exam NEW Added rationales to the comprehensive practice exams on Evolve provide step by step guidance to help you learn from your answer choices Comprehensive practice exams with over 200 questions on the Evolve companion website Sample questions in all sections prepare you for the questions you will find on the A2 Exam Step by step explanations and sample problems in the math section show you how to work through each and know how to answer Expanded HESI Hints boxes offer valuable test taking tips as well as rationales suggestions examples and reminders for specific topics A 25 question pre test at the beginning of the text helps assess areas of strength and weakness before using the text A 50 question comprehensive post test at the back of the text includes rationales for correct and incorrect answers **Strategies and Solutions to Advanced Organic Reaction Mechanisms** Andrei

Hent, John Andraos, 2019-06-26 *Strategies and Solutions to Advanced Organic Reaction Mechanisms A New Perspective on McKillop's Problems* builds upon Alexander Sandy McKillop's popular text *Solutions to McKillop's Advanced Problems in Organic Reaction Mechanisms* providing a unified methodological approach to dealing with problems of organic reaction mechanism. This unique book outlines the logic, experimental insight and problem solving strategy approaches available when dealing with problems of organic reaction mechanism. These valuable methods emphasize a structured and widely applicable approach relevant for both students and experts in the field. By using the methods described, advanced students and researchers alike will be able to tackle problems in organic reaction mechanism from the simple and straight forward to the advanced.

Development of Learning Strategies Within Chemical Education Dr. Terence McIvor, 2015-02-23 The current status of chemistry practical education and laboratory experience for the first year students is not as encouraging and useful as to produce excellent results. Consequently in this context it needs modification to implement a capstone addressing the category on development and testing of chemistry courses or learning units and demonstrating an approach to implementing a multiweek capstone project as a part of the laboratory curriculum.

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CBSE - ICSE Chemistry Part I Chandan Sengupta, First Publication October 2021. Place of Publication Arabinda Nagar Bankura 722101. This workbook will provide an ample scope in getting exposed to the system of acquiring skills and competence related to the understanding of chemistry. It also exposes the student to the concepts of chemistry for enabling the aspirant in acquisition of skills related to chemistry. Some of the worksheets are prepared along with model answers. Some other worksheets are meant for self assessment and evaluation purposes. It is also observed that some of the topics are specific to the referred curriculum. Some other topics are varyingly incorporated in other streams of study. Culmination of more than two streams will enable the fellow student to cope up with the preparatory works meant for Olympiads and other competitive examinations.

A Laboratory Outline of General Chemistry Alexander Smith, 1899 *ISC Most Likely Question Bank Chemistry Class 12 (2022 Exam) - Categorywise & Chapterwise Topics with Latest Reduced Syllabus, Answering Tips & Mind Maps* Oswal, 2021-06-15 Build your self confidence while preparing from Categorywise Chapterwise Most Likely Question Bank Series for Class 12 ISC.

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