

1 Complete the word equations for the reactions of the Group 7 elements with metals.

- a iron + chlorine → iron chloride
 b iron + bromine → iron bromide
 c sodium + iodine → sodium iodide
 d potassium + bromine → potassium bromide
 e lithium + chlorine → lithium chloride
 f sodium + chlorine → sodium chloride

2 Make up four sentences using the phrases in the table. Each sentence must include a phrase from each column.

There are	weak forces	between the two atoms of a chlorine molecule.
	strong forces	between a chlorine molecule and its neighbours.
		between the two atoms of an iodine molecule.
		between an iodine molecule and its neighbours.

*There are weak forces between the two atoms of a chlorine molecule.
 There are strong forces between a chlorine molecule and its neighbours.
 There are weak forces between the two atoms of an iodine molecule.
 There are strong forces between an iodine molecule and its neighbours.*

3 For the sentences A–H below:

- Write **1** next to the sentences that are true for Group 1.
 - Write **7** next to the sentences that are true for Group 7.
 - Write **B** next to the sentences that are true for both Groups 1 and 7.
- A Going down this group, boiling point increases.
 B Going up this group, the reactions of the elements get more vigorous.
 C Going down this group, the number of protons in atoms of the elements increases.
 D Atoms of the elements in this group form molecules made up of two atoms.
 E Going up this group, the colours of the elements get lighter.
 F The elements at the top of this group are gases.
 G The elements in this group conduct electricity.
 H All the atoms of the elements in this group have the same number of electrons in the outermost shell.

7
1
B
1
B
7
B

12 Write **M** next to the pair of elements that you predict will react most vigorously.
 Write **L** next to the pair of elements that you predict will react least vigorously.

- a sodium and iodine
 b lithium and iodine
 c potassium and fluorine
 d potassium and bromine
 e sodium and bromine

L
L
M
M
L

Answers For A Ls Chemistry

WJ Hussar



Answers For A Ls Chemistry:

Answers For A Ls Chemistry Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Answers For A Ls Chemistry**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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