

4

- 8 Methane and steam react in the presence of a catalyst.



0.5 mol of methane reacts completely with 0.5 mol of steam.

What is the volume of carbon monoxide and hydrogen produced, measured at room temperature and pressure?

	volume of CO/dm ³	volume of H ₂ /dm ³
A	0.5	1.5
B	1.0	3.0
C	12.0	12.0
D	12.0	36.0

$$n = \frac{m}{M}$$

$$n = \frac{V}{24}$$

$$0.5 \times 24 = 12 \text{ dm}^3$$

$$1.5 \times 24 = 36$$

- 9 A compound of element X has the formula X_2O and a relative formula mass of 144.

What is element X?

- A copper, Cu
B gadolinium, Gd
C sulfur, S
D tellurium, Te

1 -

or

14

$$144 - 16 = 128$$

$$128 = \text{X}_2$$

$$\frac{128}{2} = \text{X} = 64$$

5

- 10 The diagram shows the electrolysis of concentrated hydrochloric acid and concentrated aqueous sodium chloride using carbon electrodes.



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