

Name _____ Period _____

Lab – Alien Periodic Table

You are a part of a collection of scientists who have been chosen to assist a group of alien scientists. In order to be able to converse scientifically, you must learn their language, and most importantly, you must arrange their elements according to the trends that exist in the periodic table. Below are clues for the alien's elements. So far, the aliens have only discovered elements in groups 1, 2, and 13-18, and periods 1-5. Although the names of the elements are different, they must correspond to our elements if our belief of universal elements holds true. Read each clue carefully, and then place the symbol for that clue's element in the blank periodic table provided.

Alien Periodic Table

	1						18
1							
2		2					
3							
4							
5							

Alien Element Descriptions

- The inert gases are Bombal (Bo), Wobble (Wo), Jeptum (J), and Logon (L). Bombal (Bo) is a noble gas but does not have 8 valence electrons. The outside energy level of Logon (L) is its second energy level. Of these noble gases, Wobble (Wo) has the greatest atomic mass.
- The alkali metals are Xtalt (X), Byyou (By), Chow (Ch), and Quackzil (Q). Of these alkali metals, Chow (Ch) has the lowest atomic mass. Quackzil (Q) is in the same period as Wobble (Wo).
- The halogens are Apstrom (A), Vulcania (V), and Kratt (Kt). Vulcania (V) is in the same period as Quackzil (Q) and Wobble (Wo).
- The metalloids are Ernst (E), Highho (Hi), Terriblum (T), and Sississ (Ss). Sississ (Ss) is the metalloid with the highest atomic mass. Ernst (E) is the metalloid with the lowest atomic mass. Highho (Hi) and Terriblum (T) are in group 14. T has more protons than Hi. The element Yizzer (Yz) is a metalloid by location, but has properties that suggest it is a light metal.
- The most metallic element on the planet is called Xtalt (X). The most chemically active nonmetal on the planet is called Apstrom (A). The lightest element on the planet is called Pfisst (Pf). The heaviest element on the planet is Elrado (El) and it is radioactive.

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Laboratory Experiments for General Chemistry Harold R. Hunt, Toby F. Block, 1994 **General Chemistry**
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Table of Contents Alien Periodic Table Skills Lab Answer Key

1. Understanding the eBook Alien Periodic Table Skills Lab Answer Key
 - The Rise of Digital Reading Alien Periodic Table Skills Lab Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Alien Periodic Table Skills Lab Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Alien Periodic Table Skills Lab Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Alien Periodic Table Skills Lab Answer Key
 - Personalized Recommendations
 - Alien Periodic Table Skills Lab Answer Key User Reviews and Ratings
 - Alien Periodic Table Skills Lab Answer Key and Bestseller Lists
5. Accessing Alien Periodic Table Skills Lab Answer Key Free and Paid eBooks
 - Alien Periodic Table Skills Lab Answer Key Public Domain eBooks
 - Alien Periodic Table Skills Lab Answer Key eBook Subscription Services
 - Alien Periodic Table Skills Lab Answer Key Budget-Friendly Options

6. Navigating Alien Periodic Table Skills Lab Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Alien Periodic Table Skills Lab Answer Key Compatibility with Devices
 - Alien Periodic Table Skills Lab Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Alien Periodic Table Skills Lab Answer Key
 - Highlighting and Note-Taking Alien Periodic Table Skills Lab Answer Key
 - Interactive Elements Alien Periodic Table Skills Lab Answer Key
8. Staying Engaged with Alien Periodic Table Skills Lab Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Alien Periodic Table Skills Lab Answer Key
9. Balancing eBooks and Physical Books Alien Periodic Table Skills Lab Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Alien Periodic Table Skills Lab Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Alien Periodic Table Skills Lab Answer Key
 - Setting Reading Goals Alien Periodic Table Skills Lab Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Alien Periodic Table Skills Lab Answer Key
 - Fact-Checking eBook Content of Alien Periodic Table Skills Lab Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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web section 1 answer the following questions in the space provided 1 c in the modern periodic table elements are ordered a according to decreasing atomic mass b according to mendeleev s original design c according to increasing atomic number d based on when they were discovered 2 d

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web the periodic law states that there is periodic repetition of the physical and chemical properties of elements when they are arranged in order of increasing atomic mass if only metals are considered when they are arranged in order of increasing atomic radii

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web compare the radius of a negative ion to the radius of its neutral atom the radius of a negative ion is larger than the radius of the neutral atom give the approximate positions and blocks where metals and nonmetals are found in the periodic table

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web a the d sub orbital can hold ten electrons making the d block ten elements wide b the d block is the most researched area of the periodic table c the elements in the d block are more important than the elements in the rest of the table

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