Spectral Analysis



Types of spectra:

Type of spectrum	Photographic example
Continuous (or continuum)	
Absorption (dark line)	
Emission (bright line)	

Here is a spectrum with many of the known spectral lines.



Designation	Element	Wavelength (nm)	Designation	Element	Warningth (nm)
,	0.	998,765	34 3	Te .	495,763
Z	O ₁	822,696	P	HB	486.134
A	0,	759.379	34	76	466.834
8	O ₁	686,719		Z+	438.335
C X	He	656.281	0	Ha	434.047
	0:	627,661	G	Zer .	430,790
D_{ij}	No.	589.592	G		430,734
D ₁	Na	588.995	The second	154	430.175
Dr or d	He.	587,5688	H	Cal	396.547
• 7	He	546.073	K	Cal	295,368
Le	2	527,009	L	76	382.044
be .	Ma	518,342	N	Ze:	358.121
600 E	Mg	517.276	P		356,332
lhe .	Fe .	516.891	T	Fe:	302.108
be a	Per	516.891	(A)	100	299.444
he .	Mg	556,755			

What colour is a	photon of light wi	th wavelength	
a. 450 nm	6	700 nm	
c. 523 nm		560 nm	
. What colour is t	he emission of hyd	irogen-alpha, H g	2
Drow the main o	pectrum lines for a	stor contoinino	Trop and Hudinas
(note: Some eler	nents have more ti	non one obsorptio	on line)
- Name of the last			
400	500	4000	700
400	The second secon		
400	The second secon	fs (nanometers)	10.70
- 	Wavdoug	th (nanonestern)	NOTE:
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Analysis Of Spectral Lines Answers

Helge Holst, Hendrik Anthony Kramers

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