

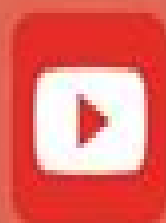
IGCSE A* MODEL ANSWERS

0620 CHEMISTRY

MAY/JUNE - 2024

PAPER - 61

0620_S24_QP_61 (Q3-4)



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Table 1 shows the tests and the student's observations for solid P.

Tests	Observations
Test 1 Heat about half of solid P in boiling tube with test tube in fumes of HCl .	<u>Amorphous white</u> The white solid forms a colourless liquid, steam effervesces from the boiling tube and condensation is seen at the top of the boiling tube. After a while the colourless liquid becomes a white solid.
Test 2 The remaining solid P is dissolved in water to form solution P. Solution P is divided into three portions. To the first portion of solution P, a glowing splint, with a glowing <u>blue</u> flame, is held over the mouth of the boiling tube and then is removed. Repeat the procedure and then bring red litmus paper at the mouth of the boiling tube.	<u>no H_2 gas</u> <u>no H_2S gas</u> <u>no H_2SO_4</u> <u>no change</u> The glowing red litmus paper remains red.
Test 3 To the second portion of solution P, add 1 cm ³ of dilute <u>nitric acid</u> followed by a few drops of sodium <u>chromate(VI)</u> solution.	<u>white precipitate</u> SO_4^{2-}
Test 4 To the third portion of solution P, add a few drops of acidified aqueous potassium manganate(VII).	The solution becomes light purple. SO_3^{2-}

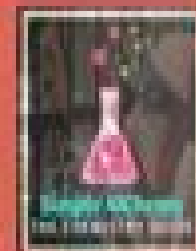
10. The observations in Test 1 show that solid P is hydrated.

11. Suggest a chemical test to show that the condensation at the top of the boiling tube contains water.

— anhydrous blue cobalt chloride paper

— pink H_2O

— anhydrous white crumbly solid ✓



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