

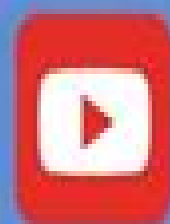
IGCSE A* MODEL ANSWERS

0620 CHEMISTRY

MAY/JUNE - 2024

PAPER - 62

0620_S24_QP_62 (Q1-2)



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When a mixture of calcium hydroxide and ammonium chloride is heated, ammonia gas is made.

$\text{calcium hydroxide} + \text{ammonium chloride} \rightarrow \text{calcium chloride} + \text{water} + \text{ammonia}$ (1/1)

Ammonia gas is soluble in water and toxic. Ammonia gas reacts with hot copper(II) oxide to make

$\text{ammonia} + \text{copper(II) oxide} \rightarrow \text{copper} + \text{water} + \text{nitrogen}$

A student makes nitrogen using the apparatus shown in Fig. 1.1.

calcium hydroxide and ammonium chloride



Fig. 1.1

(a) Name the parts of apparatus labelled A and B in Fig. 1.1.

A. delivery tube

B. inverting cylinder

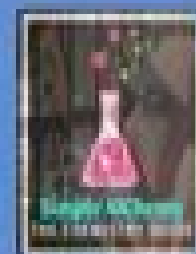
(b) The apparatus needs to be heated in two places.

On Fig. 1.1, draw arrows in two places to show where the apparatus should be heated during the experiment.

(c) During the reaction, a colourless liquid collects at the point marked (i) in Fig. 1.1.

Suggest the identity of the colourless liquid.

Water



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