

- 4 A student investigated the reaction between aqueous potassium manganate(VII), which is purple, and two different colourless acidic solutions, **D** and **E**.

For Examiner's Use

Three experiments were carried out.

(a) Experiment 1

A burette was filled with the solution of potassium manganate(VII) to the 0.0 cm³ mark. Using a measuring cylinder, 25 cm³ of solution **D** was poured into a conical flask.

Potassium manganate(VII) solution was added to the flask until the mixture just turned permanently pink.

Use the burette diagram to record the final volume in the table and complete the table.



final reading

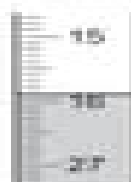
	burette reading
final burette reading / cm ³	
initial burette reading / cm ³	
difference / cm ³	

[2]

(b) Experiment 2

Experiment 1 was repeated using 25 cm³ of solution **E** instead of solution **D**.

Use the burette diagrams to record the readings in the table and complete the table.



initial reading



final reading

	burette reading
final burette reading / cm ³	
initial burette reading / cm ³	
difference / cm ³	

[2]

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Table of Contents **0620 63 O N 13 Markscheme**

1. Understanding the eBook 0620 63 O N 13 Markscheme
 - The Rise of Digital Reading 0620 63 O N 13 Markscheme
 - Advantages of eBooks Over Traditional Books
2. Identifying 0620 63 O N 13 Markscheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 0620 63 O N 13 Markscheme
 - User-Friendly Interface
4. Exploring eBook Recommendations from 0620 63 O N 13 Markscheme
 - Personalized Recommendations
 - 0620 63 O N 13 Markscheme User Reviews and Ratings
 - 0620 63 O N 13 Markscheme and Bestseller Lists

5. Accessing 0620 63 O N 13 Markscheme Free and Paid eBooks
 - 0620 63 O N 13 Markscheme Public Domain eBooks
 - 0620 63 O N 13 Markscheme eBook Subscription Services
 - 0620 63 O N 13 Markscheme Budget-Friendly Options
6. Navigating 0620 63 O N 13 Markscheme eBook Formats
 - ePub, PDF, MOBI, and More
 - 0620 63 O N 13 Markscheme Compatibility with Devices
 - 0620 63 O N 13 Markscheme Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 0620 63 O N 13 Markscheme
 - Highlighting and Note-Taking 0620 63 O N 13 Markscheme
 - Interactive Elements 0620 63 O N 13 Markscheme
8. Staying Engaged with 0620 63 O N 13 Markscheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 0620 63 O N 13 Markscheme
9. Balancing eBooks and Physical Books 0620 63 O N 13 Markscheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 0620 63 O N 13 Markscheme
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 0620 63 O N 13 Markscheme
 - Setting Reading Goals 0620 63 O N 13 Markscheme
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 0620 63 O N 13 Markscheme
 - Fact-Checking eBook Content of 0620 63 O N 13 Markscheme
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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