

## 4.2 Solutions – Solubility Curves

Indicate the following regions (for your convenience):

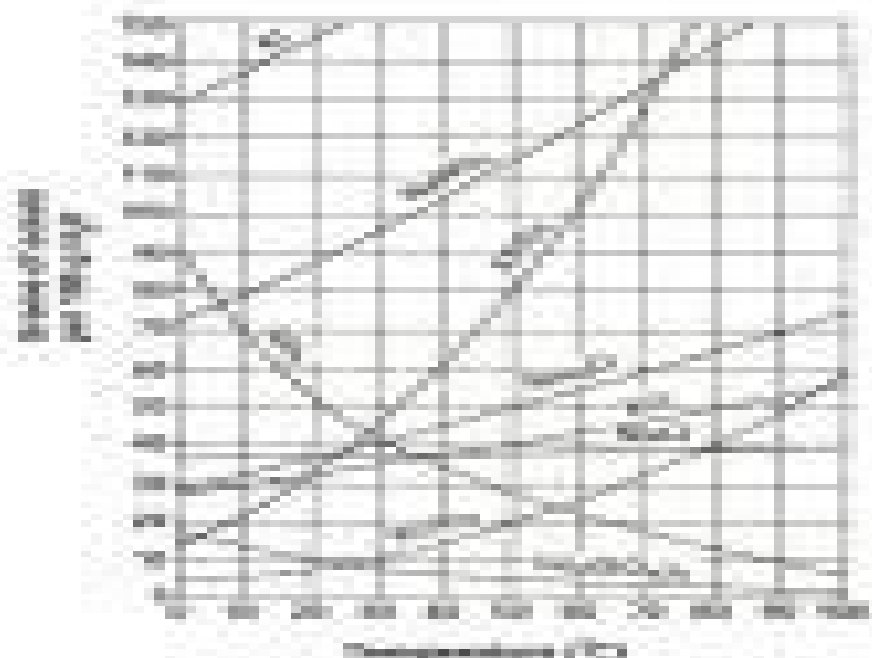
**Saturated solution:** Region immediately above solubility curve (no undissolved solute with stirring)

**Unsaturated solution:** Below solubility curve (all solute dissolved with no excess solute present)

**Supersaturated solution:** Below solubility curve to right of the maximum solubility point (if cooled without stirring)

Indicate the following quantities based on the solubility curve provided:

1. Dissolving 40 g of  $\text{KCl}$  in 100 g of water at  $20^\circ\text{C}$   
**Unsaturated solution**
2. Mass of  $\text{KCl}$  dissolved in 100 g of water at  $40^\circ\text{C}$   
**Saturated solution**
3. Mass of  $\text{KCl}$  dissolved in 40 g of water at  $40^\circ\text{C}$   
**Saturated solution**
4. A saturated solution contains 40 g of  $\text{KCl}$  in 100 g of water at  $40^\circ\text{C}$ . Is this solution saturated?
5. What mass of  $\text{KCl}$  is dissolved in 100 g of water at  $20^\circ\text{C}$ ?
6. A saturated solution contains 40 g of  $\text{KCl}$  in 100 g of water at  $40^\circ\text{C}$ . What mass of  $\text{KCl}$  is dissolved in 100 g of water at  $20^\circ\text{C}$ ?
7. A saturated solution contains 40 g of  $\text{KCl}$  in 100 g of water at  $40^\circ\text{C}$ . What mass of  $\text{KCl}$  is dissolved in 100 g of water at  $20^\circ\text{C}$ ?
8. A saturated solution contains 40 g of  $\text{KCl}$  in 100 g of water at  $40^\circ\text{C}$ . What mass of  $\text{KCl}$  is dissolved in 100 g of water at  $20^\circ\text{C}$ ?
9. A saturated solution contains 40 g of  $\text{KCl}$  in 100 g of water at  $40^\circ\text{C}$ . What mass of  $\text{KCl}$  is dissolved in 100 g of water at  $20^\circ\text{C}$ ?
10. A saturated solution contains 40 g of  $\text{KCl}$  in 100 g of water at  $40^\circ\text{C}$ . What mass of  $\text{KCl}$  is dissolved in 100 g of water at  $20^\circ\text{C}$ ?
11. A saturated solution contains 40 g of  $\text{KCl}$  in 100 g of water at  $40^\circ\text{C}$ . What mass of  $\text{KCl}$  is dissolved in 100 g of water at  $20^\circ\text{C}$ ?
12. A saturated solution contains 40 g of  $\text{KCl}$  in 100 g of water at  $40^\circ\text{C}$ . What mass of  $\text{KCl}$  is dissolved in 100 g of water at  $20^\circ\text{C}$ ?
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17. A saturated solution contains 40 g of  $\text{KCl}$  in 100 g of water at  $40^\circ\text{C}$ . What mass of  $\text{KCl}$  is dissolved in 100 g of water at  $20^\circ\text{C}$ ?
18. A saturated solution contains 40 g of  $\text{KCl}$  in 100 g of water at  $40^\circ\text{C}$ . What mass of  $\text{KCl}$  is dissolved in 100 g of water at  $20^\circ\text{C}$ ?
19. A saturated solution contains 40 g of  $\text{KCl}$  in 100 g of water at  $40^\circ\text{C}$ . What mass of  $\text{KCl}$  is dissolved in 100 g of water at  $20^\circ\text{C}$ ?
20. A saturated solution contains 40 g of  $\text{KCl}$  in 100 g of water at  $40^\circ\text{C}$ . What mass of  $\text{KCl}$  is dissolved in 100 g of water at  $20^\circ\text{C}$ ?



# 1 Solutions Solubility Curves Key

**Peter Ellis, Derek McMonagle**



## 1 Solutions Solubility Curves Key:

**Ascent! 1** Louise Petheram, Phil Routledge, Lawrie Ryan, 2002 This series is focused on delivering custom materials which are designed and presented to meet the needs of enthusiastic and committed students The resources are written at an average reading ability level but with full and proper use of scientific terminology throughout Ascent has its own text linked website [www.nelsonthornes.com/ascent](http://www.nelsonthornes.com/ascent) **Class 8-12 Chemistry Questions and Answers PDF** Arshad Iqbal, The Class 8 12 Chemistry Quiz Questions and Answers PDF Grade 8 12 Chemistry Competitive Exam Questions Chapter 1 15 Practice Tests Chemistry Textbook Questions for Beginners includes Questions to solve problems with hundreds of class questions Class 8 12 Chemistry Questions and Answers PDF book covers basic concepts and analytical assessment tests Class 8 12 Chemistry Quiz PDF book helps to practice test questions from exam prep notes The Grade 8 12 Chemistry Quiz Questions and Answers PDF eBook includes Practice material with verbal quantitative and analytical past papers questions Class 8 12 Chemistry Questions and Answers PDF Free download chapter 1 a book to review textbook questions on chapters Molecular structure acids and bases atomic structure bonding chemical equations descriptive chemistry equilibrium systems gases laboratory liquids and solids mole concept oxidation reduction rates of reactions solutions thermochemistry Questions for high school and college revision questions Chemistry Interview Questions and Answers PDF Download free eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The Grade 8 12 Chemistry Interview Questions Chapter 1 15 PDF book includes high school workbook questions to practice Questions for exam Chemistry Practice Tests a textbook s revision guide with chapters Questions for NEET MCAT GRE GMAT SAT ACT competitive exam Grade 8 12 Chemistry Questions Bank Chapter 1 15 PDF book covers problem solving exam tests from chemistry practical and textbook s chapters as Chapter 1 Molecular Structure Questions Chapter 2 Acids and Bases Questions Chapter 3 Atomic Structure Questions Chapter 4 Bonding Questions Chapter 5 Chemical Equations Questions Chapter 6 Descriptive Chemistry Questions Chapter 7 Equilibrium Systems Questions Chapter 8 Gases Questions Chapter 9 Laboratory Questions Chapter 10 Liquids and Solids Questions Chapter 11 Mole Concept Questions Chapter 12 Oxidation Reduction Questions Chapter 13 Rates of Reactions Questions Chapter 14 Solutions Questions Chapter 15 Thermochemistry Questions The Molecular Structure Quiz Questions PDF e Book Chapter 1 interview questions and answers on polarity three dimensional molecular shapes The Acids and Bases Quiz Questions PDF e Book Chapter 2 interview questions and answers on Arrhenius concept Bronsted lowry concept indicators introduction Lewis concept pH strong and weak acids and bases The Atomic Structure Quiz Questions PDF e Book Chapter 3 interview questions and answers on electron configurations experimental evidence of atomic structure periodic trends quantum numbers and energy levels The Bonding Quiz Questions PDF e Book Chapter 4 interview questions and answers on ionic bond covalent bond dipole dipole forces hydrogen bonding intermolecular forces London dispersion forces metallic bond The Chemical Equations Quiz Questions PDF e Book Chapter 5 interview questions

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Chemistry for OCR A for Separate Award David Lees, John Payne, 2001 This series is for schools following OCR A double or separate award for GCSE science The resources offer preparation for the OCR exams with teacher support to minimise time spent on administration The teacher s resources are available on CD ROM in a fully customizable format

Scientifica Essentials Peter Ellis, Derek McMonagle, 2004 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities

*Regents Exams and Answers: Chemistry--Physical Setting Revised Edition* Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron s Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day

*Scientifica Pupil Book 7 (Levels 4-7)* David Sang, Peter Ellis, Derek McMonagle, 2004 This student book covers Levels 4 7 and is structured to match the sequence of the QCA Scheme of Work

Units and the National Framework for Science Guidelines Each lesson can commence with a really quick starter activity The teacher support materials of course provide hundreds more Scientific aims to provide just the right proportion of reading versus doing There is enough text on each page for students to develop their literacy skills but each lesson spread also contains an optional activity or two to access the real experience of Science Ideas and Evidence articles are presented in each text in a more magazine style

43 Years Chapterwise Topicwise Solved Papers (2021-1979) IIT JEE Chemistry  
 Ranjeet Shahi, 2021-01-11

1 43 Years Chapterwise and Topicwise Solved papers for JEE Main Advanced 2 The book is divided into 33 Chapters 3 Ample Questions are given 2021 1979 for practice 4 JEE Advanced Solved Papers are provided to know the paper pattern Cracking one of the toughest examinations requires great deal of determination and efforts from the students that can only be achieved from the previous year's solved papers that provide complete idea of types of questions asked and pattern of paper Prepared under the observation of the subject expert the updated edition of 43 years Chapterwise Topicwise Solved Papers 2021 1979 of Chemistry is a one stop solution for the preparation of IIT JEE Mains and Advanced Giving complete coverage to the syllabus this book has been categorized under 33 chapters that are supplemented with good number of questions of both JEE Mains and Advanced in Chapterwise and Topicwise manner For further practice Previous Years Solved Papers and Selected Questions of JEE Main Jan Sept 2021 are given at the end of the book to help aspirants for the forthcoming exam Table of Content Some Basic Concepts of Chemistry Atomic Structure Periodic Classification and Periodic Properties Chemical Bonding States of Matter Chemical and Ionic Equilibrium Thermodynamics and Thermochemistry Solid State Solutions and Colligative Properties Electrochemistry Chemical Kinetics Nuclear Chemistry Surface Chemistry s block Elements p block Elements I p block Elements II Transition and Inner Transition Elements Coordination Compounds Extraction of Metals Qualitative Analysis Organic Chemistry Basics Hydrocarbons Alkyl Halides Alcohols and Ethers Aldehydes and Ketones Carboxylic Acids and their Derivatives Aliphatic Compounds Containing Nitrogen Benzene and Alkyl Benzene Aromatic Compounds Containing Nitrogen Aryl Halides and Phenols Aromatic Aldehydes Ketones and Acids Biomolecules and Chemistry in Everyday Life Environmental Chemistry JEE Advanced Solved Paper 2021

*Spotlight Science* Keith Johnson, Sue Adamson, Gareth Williams, 2002 This Spiral Edition Teacher Support Pack offers comprehensive support and guidance providing the best possible learning experience for your students and saving time for everyone in the department

*Clinical Aspects of Albumin* S.H. Yap, C.L.H. Majoor, J.H.M. van Tongeren, 2012-12-06 Albumin is the most abundant serum protein produced by the liver In clinical practice the serum level of albumin continues to be used as an important marker of the presence progress or of the improvement of many diseases even though it is the complex end result of synthesis degradation and distribution between intra and extravascular space The clinical history of albumin began as early as in 1837 when Ansell first recognized albumen and noted that this protein is needed for transport functions for maintaining fluidity of the vascular system and for the prevention of edema However the important physiological properties

of serum proteins and their role in the regulation of the oncotic pressure were demonstrated later by the physiologist E H Starling in 1895 In 1917 the clinician A A Epstein first described the edema in patients with the nephrotic syndrome as being a result of a very low level of serum albumin Although the determination of serum albumin concentration became more popular after Howe in 1921 introduced the technique of separation of serum globulins from albumin by sodium sulfate the first preparations of human serum albumin were made available for clinical use in only 1941 by the development of plasma fractionation by Cohn and his coworkers at Harvard Medical School

*Handbook of Industrial Crystallization* Allan Myerson, 2002-01-08 Crystallization is an important separation and purification process used in industries ranging from bulk commodity chemicals to specialty chemicals and pharmaceuticals In recent years a number of environmental applications have also come to rely on crystallization in waste treatment and recycling processes The authors provide an introduction to the field of newcomers and a reference to those involved in the various aspects of industrial crystallization It is a complete volume covering all aspects of industrial crystallization including material related to both fundamentals and applications This new edition presents detailed material on crystallization of biomolecules precipitation impurity crystal interactions solubility and design Provides an ideal introduction for industrial crystallization newcomers Serves as a worthwhile reference to anyone involved in the field Covers all aspects of industrial crystallization in a single complete volume

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