

Name: _____

Chemistry Practice: Balancing Chemical Equations

1. Balance the equations below. Remember this means only using coefficients. All formulas are written correctly.

- a.) $2 \text{NaHCO}_3 \longrightarrow \text{Na}_2\text{CO}_3 + \text{H}_2\text{O} + \text{CO}_2$
- b.) $\text{Zn} + \text{S} \longrightarrow \text{ZnS}$
- c.) $2 \text{H}_2 + \text{O}_2 \longrightarrow 2 \text{H}_2\text{O}$
- d.) $2 \text{FeO} + \text{C} \longrightarrow 2 \text{Fe} + \text{CO}_2$
- e.) $\text{Cl}_2 + 2 \text{KI} \longrightarrow 2 \text{KCl} + \text{I}_2$
- f.) $\text{B}_2\text{O}_3 + 3 \text{Mg} \longrightarrow 3 \text{MgO} + 2 \text{B}$
- g.) $\text{Ca} + 2 \text{H}_2\text{O} \longrightarrow \text{Ca(OH)}_2 + 2 \text{H}_2$
- h.) $4 \text{H}_2\text{O} + \text{Na}_2\text{B}_4\text{O}_7 + \text{H}_2\text{SO}_4 \longrightarrow 4 \text{H}_3\text{BO}_3 + \text{Na}_2\text{SO}_4$
- i.) $\text{CaC}_2 + 2 \text{H}_2\text{O} \longrightarrow \text{Ca(OH)}_2 + \text{C}_2\text{H}_2$
- j.) $\text{NH}_4\text{NO}_3 \longrightarrow \text{N}_2\text{O} + 2 \text{H}_2\text{O}$
- k.) $\text{H}_2\text{S} + \text{O}_2 \longrightarrow \text{SO}_2 + \text{H}_2\text{O}$
- l.) $\text{H}_2\text{SO}_4 + 2 \text{NaCl} \longrightarrow 2 \text{HCl} + \text{Na}_2\text{SO}_4$
- m.) $2 \text{NaCl} + 2 \text{H}_2\text{O} \longrightarrow 2 \text{NaOH} + \text{Cl}_2 + \text{H}_2$
- n.) $3 \text{BaO} + 2 \text{Al} \longrightarrow 3 \text{Ba} + \text{Al}_2\text{O}_3$
- o.) $3 \text{O}_2 \longrightarrow 2 \text{O}_3$
- p.) $\text{SiO}_2 + 2 \text{C} \longrightarrow 2 \text{CO} + \text{Si}$
- q.) $2 \text{PbS} + 3 \text{O}_2 \longrightarrow 2 \text{PbO} + 2 \text{SO}_2$
- r.) $2 \text{HgO} \longrightarrow 2 \text{Hg} + \text{O}_2$
- s.) $\text{C}_{12}\text{H}_{22}\text{O}_{11} + 12 \text{O}_2 \longrightarrow 12 \text{CO}_2 + 11 \text{H}_2\text{O}$
- t.) $\text{C}_3\text{H}_8 + 5 \text{O}_2 \longrightarrow 3 \text{CO}_2 + 4 \text{H}_2\text{O}$
- u.) $2 \text{FeCl}_3 + 3 (\text{NH}_4)_2\text{S} \longrightarrow 6 \text{NH}_4\text{Cl} + \text{Fe}_2\text{S}_3$
- v.) $\text{P}_2\text{O}_5 + 6 \text{NaOH} \longrightarrow 2 \text{Na}_3\text{PO}_4 + 3 \text{H}_2\text{O}$
- w.) $3 \text{BaCl}_2 + \text{Al}_2(\text{SO}_4)_3 \longrightarrow 3 \text{BaSO}_4 + 2 \text{AlCl}_3$
- x.) $4 \text{NH}_3 + 5 \text{O}_2 \longrightarrow 6 \text{H}_2\text{O} + 4 \text{NO}$
- y.) $2 \text{Na}_2\text{O}_2 + 2 \text{H}_2\text{O} \longrightarrow 4 \text{NaOH} + \text{O}_2$
- z.) $\text{Sn(ClO}_3)_4 \longrightarrow \text{SnCl}_4 + 6 \text{O}_2$

Text Problems to be Completed on the Other Side of this Worksheet:

*P. 173 #45,49,53,54,56,57,58,63, 65,66,67,71

*(Although the directions may say to write an "unbalanced" equation, write a balanced equation for each.)

10 1 Reactions And Equations Answer Key

Frankline E. Niles, Edna L. Lortie



10 1 Reactions And Equations Answer Key:

Solutions of Reaction Rate Equations Pertaining to Electron Irradiation of 4:1 Mixtures of N₂ and O₂

Frankline E. Niles, Edna L. Lortie, 1967 One of the Keneshea computer codes see AD 424 173 was adapted for use on the Ballistic Research Laboratories Electronic Scientific Computer Using this modified code reaction rate equations were solved for the following 15 species e NO₂ O O₂ O₃ N₂ NO O O₂ N NO N₂O NO₂ O and O₃ The calculations were made for a 4 : 1 mixture of N₂ and O₂ at 1 torr total pressure and 300K Rate constants as given by Keneshea and Fowler see AD 646 975 were used The solutions are presented as number densities versus time after the start of the irradiating electron beam A description of the modified code is presented Author **CBSE Chapterwise Worksheets for Class 10** Gurukul, 2021-07-30 Practice Perfectly and Enhance Your CBSE Class 10th Board preparation with Gurukul's CBSE Chapterwise Worksheets for 2022 Examinations Our Practicebook is categorized chapterwise topicwise to provide you in depth knowledge of different concept topics and questions based on their weightage to help you perform better in the 2022 Examinations How can you Benefit from CBSE Chapterwise Worksheets for 10th Class 1 Strictly Based on the Latest Syllabus issued by CBSE 2 Includes Checkpoints basically Benchmarks for better Self Evaluation for every chapter 3 Major Subjects covered such as Science Mathematics Social Science 4 Extensive Practice with Assertion Reason Case Based MCQs Source Based Questions 5 Comprehensive Coverage of the Entire Syllabus by Experts Our Chapterwise Worksheets include Mark Yourself at the end of each worksheet where students can check their own score and provide feedback for the same Also consists of numerous tips and tools to improve problem solving techniques for any exam paper Our book can also help in providing a comprehensive overview of important topics in each subject making it easier for students to solve for the exams *Numerical Solution of Time-Dependent Advection-Diffusion-Reaction Equations* Willem Hundsdorfer, Jan G. Verwer, 2013-04-17 This book deals with numerical methods for solving partial differential equations PDEs coupling advection diffusion and reaction terms with a focus on time dependency A combined treatment is presented of methods for hyperbolic problems thereby emphasizing the one way wave equation methods for parabolic problems and methods for stiff and non stiff ordinary differential equations ODEs With regard to time dependency we have attempted to present the algorithms and the discussion of their properties for the three different types of differential equations in a unified way by using semi discretizations i.e the method of lines whereby the PDE is transformed into an ODE by a suitable spatial discretization In addition for hyperbolic problems we also discuss discretizations that use information based on characteristics Due to this combination of methods this book differs substantially from more specialized textbooks that deal exclusively with numerical methods for either PDEs or ODEs We treat integration methods suitable for both classes of problems This combined treatment offers a clear advantage On the one hand in the field of numerical ODEs highly valuable methods and results exist which are of practical use for solving time dependent PDEs something which is often not fully exploited by numerical PDE researchers Although many problems can be

solved by Euler's method or the Crank Nicolson method better alternatives are often available which can significantly reduce the computational effort needed to solve practical problems

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Calculus for the Life Sciences: A Modeling Approach James L. Cornette, Ralph A. Ackerman, 2019-05-25 Calculus for the Life Sciences is an entire reimagining of the standard calculus sequence with the needs of life science students as the fundamental organizing principle Those needs according to the National Academy of Science include the mathematical concepts of change modeling equilibria and stability structure of a system interactions among components data and measurement visualization and algorithms This book addresses in a deep and significant way every concept on that list The book begins with a primer on modeling in the biological realm and biological modeling is the theme and frame for the entire book The authors build models of bacterial growth light penetration through a column of water and dynamics of a colony of mold in the first few pages In each case there is actual data that needs fitting In the case of the mold colony that data is a set of photographs of the colony growing on a ruled sheet of graph paper and the students need to make their own approximations Fundamental questions about the nature of mathematical modeling trying to approximate a real world phenomenon with an equation are all laid out for the students to wrestle with The authors have produced a beautifully written introduction to the uses of mathematics in the life sciences The exposition is crystalline the problems are overwhelmingly from biology and interesting and rich and the emphasis on modeling is pervasive An instructor's manual for this title is available electronically to those instructors who have adopted the textbook for classroom use Please send email to textbooks@ams.org for more information Online question content and interactive step by step tutorials are available for this title in WebAssign WebAssign is a leading provider of online instructional tools for both faculty and students

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Linear Functions and Matrix Theory Bill Jacob, 2012-12-06

Courses that study vectors and elementary matrix theory and introduce linear transformations have proliferated greatly in recent years Most of these courses are taught at the undergraduate level as part of or adjacent to the second year calculus sequence Although many students will ultimately find the material in these courses more valuable than calculus they often experience a class that consists mostly of learning to implement a series of computational algorithms The objective of this text is to bring a different vision to this course including many of the key elements called for in current mathematics teaching reform efforts Three of the main components of this current effort are the following

- 1 Mathematical ideas should be introduced in meaningful contexts with after a clear understanding formal definitions and procedures developed of practical situations has been achieved
- 2 Every topic should be treated from different perspectives including the numerical geometric and symbolic viewpoints
- 3 The important ideas need to be visited repeatedly throughout the term with students understanding deepening each time

This text was written with these three objectives in mind The first two chapters deal with situations requiring linear functions at times locally linear functions or linear ideas in geometry for their

understanding These situations provide the context in which the formal mathematics is developed and they are returned to with increasing sophistication throughout the text **Kinetic Models of Catalytic Reactions** G.S. Yablonskii,V.I.

Bykov,V.I. Elokhin,A.N. Gorban,1991-04-17 This book has been written by a group of mathematicians and chemists whose common interest is in the complex dynamics of catalytic reactions Based on developments in mathematical chemistry a general theory is described that allows the investigation of the relationships between the kinetic characteristics of complex reactions and their detailed reaction mechanism Furthermore a comprehensive analysis is made of some typical mechanism of catalytic reactions in particular for the oxidation of carbon monoxide on platinum metals In fact the book presents three kinetics a detailed oriented to the elucidation of a detailed reaction mechanism according to its kinetic laws b applied with the aim of obtaining kinetic relationships for the further design of chemical reactors and c mathematical kinetics whose purpose is the analysis of mathematical models for heterogeneous catalytic reactions taking place under steady or unsteady state conditions

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