Chapter 10 Review: Chemical Quantities & Calculations

(aka - mole & volume calculations, % composition, empirical & molecular formulas)

For the test, be sure that you are familiar, and able to complete, the problems below as well as the problems found on the following worksheets: "Molecular Weight and Mole Calculations," "Percent Comp. & Chemical Formula Calculations" and "Chapter 10 Sample Problems." Check the class website for answers to the review worksheet.

- The air around you is made up of 20.95% oxygen. How many oxygen molecules are in 75.0 grams of air? The density of air is 1.19 g/L. assume STP.
- How much helium (grams) would be needed to inflate a balloon to a volume of 5.50L? assume STP.
- The molecular formula of an antibacterial drug is C₁₇H₁₈FN₂O₃. How many fluorine atoms are in 150mg of this drug?

4. A molecule is made up of 40.0% carbon, 6.67% hydrogen, and an unknown amount of oxygen. What is the molecular formula of the molecule if its molar mass is 90.0g/mole?

- What is the empirical formula for a compound made up of 71.72% chlorine, 16.16% oxygen and 12.12% carbon? Calculate the molecular formula for the compound if it has a molar mass of 296,7g/mol.
- 6. How many molecules are in 8.42 Kilograms of C2H4?

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David Kirk

10 Chemical Quantities Guided Practice Problems Answers 234326:

10 Chemical Quantities Guided Practice Problems Answers 234326 Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the power of words has become more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book 10 Chemical Quantities Guided Practice Problems Answers 234326, a literary masterpiece that delves deep to the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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