Mechanism Summary for A-level AQA Chemistry

Electrophilic Addition of alkenes with bromine

Electrophilic Addition of alloenes with sulfuric acid

Electrophilic Addition of alkenes with hydrogen bromide

Elimination of halogenoalkanes with ethanolic hydroxide ions

BHT.

8467

Acid catalysed elimination mechanism: alcohols → alkenes

Acid catalysed addition mechanism for hydration of ethene

$$H_3C$$
 H_3C
 H_3C

Nucleophilic Substitution of Halogenoalkanes with cyanide ions.

$$H_3C$$
 H_3C
 H_3C

Free Radical Substitution of alkanes with bromine

STEP ONE Initiation

Essential condition: UV light

Br, → 2Br *

STEP TWO Propagation

CH3CH2" + Br2 → CH3CH2Br + Br

STEP THREE Termination

CH3CH2" + Br" → CH3CH2Br

CH2CH2, + CH2CH3, → CH2CH2CH2CH3

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Jenny Frost, Tony Turner

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