# Anions, Cations, and Ionic Reactions

Hands-On Labs, Inc.
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## Lab Report Assistant

This document is not meant to be a substitute for a formal laboratory report. The Lab Report Assistant is simply a summary of the experiment's questions, diagrams if needed, and data tables that should be addressed in a formal lab report. The intent is to facilitate students' writing of lab reports by providing this information in an editable file which can be sent to an instructor.

# Exercise 1: Qualitative Anion Tests

**Data Table 1.** Anion Confirmation Tests.

| Chemical | Anion | Observations: Addition of $AgNO_3$ | Observations: Addition of HCl |
|---|---|---|---|
| **NaBr** | **Bromide** | Turned milky white | Turned yellowish<br>More of a gel |
| **$Na_2CO_3$** | **Carbonate** | Thick white glob<br>A little grainy | White sediment in a clear liquid<br>Bubbles up |
| **NaCl** | **Chloride** | Milky white | White sediment visible |
| **NaI** | **Iodide** | Yellow glob immediately when dropped in | Yellow, grainy/sediment, thick looking |
| **$Na_2SO_4$** | **Sulfate** | Thickened | Very cloudy/milky |
| **$Na_2S$** | **Sulfide** | Brown Globs immediately when dropped in<br>Still watery | Brown globs thinned out into brown pieces of sediment<br>Still watery (brownish) looks like dirty water |
| **Unknown #103** | Carbonate | Thick white glob<br>Turned gel | Bubbly and cloudy<br>Very thick gel |
| **Unknown #104** | Sulfide | Brown globs immediately when dropped in<br>Still watery | Brown globs thinned out into brown pieces of sediment<br>Still watery (brownish) looks like dirty water |

# Analysis Of Anions Cations Lab Answers

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