

CHAPTER 7

STUDY GUIDE

Ionic Compounds and Metals

Section 7.1 Ion Formation

In your textbook, read about chemical bonds and formation of ions.

Use each of the terms below just once to complete the passage.

chemical bond	electrons	energy level	ions	noble gases
nucleus	octet	pseudo-noble gas formations	valence	

The force that holds two atoms together is called a(n) (1) _____.

Such an attachment may form by the attraction of the positively charged

(2) _____ of one atom for the negatively charged

(3) _____ of another atom, or by the attraction of charged atoms, which are called

(4) _____. The attractions may also involve

(5) _____ electrons, which are the electrons in the outermost

(6) _____. The (7) _____ are a family of elements that have very

little tendency to react. Most of these elements have a set of eight outermost electrons, which is called a

stable (8) _____. The relatively stable electron structures developed by loss of

electrons in certain elements of groups 3, 4, 13, and 14 are called (9) _____.

For each statement below, write *true* or *false*.

- _____ 10. A positively charged ion is called an anion.
- _____ 11. Elements in group 1 lose their one valence electron, forming an ion with a 1+ charge.
- _____ 12. Elements tend to react so that they acquire the electron structure of a halogen.
- _____ 13. A sodium atom tends to lose one electron when it reacts.
- _____ 14. The electron structure of a zinc ion (Zn^{2+}) is an example of a pseudo-noble gas formation.
- _____ 15. A Cl^- ion is an example of a cation.
- _____ 16. The ending *-ide* is used to designate an anion.
- _____ 17. Nonmetals form a stable outer electron configuration by losing electrons and becoming anions.

Answer Study Guide Ionic Compounds And Metals

Martha Mackin



Answer Study Guide Ionic Compounds And Metals:

The book delves into Answer Study Guide Ionic Compounds And Metals. Answer Study Guide Ionic Compounds And Metals is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Answer Study Guide Ionic Compounds And Metals, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Answer Study Guide Ionic Compounds And Metals
- Chapter 2: Essential Elements of Answer Study Guide Ionic Compounds And Metals
- Chapter 3: Answer Study Guide Ionic Compounds And Metals in Everyday Life
- Chapter 4: Answer Study Guide Ionic Compounds And Metals in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Answer Study Guide Ionic Compounds And Metals. This chapter will explore what Answer Study Guide Ionic Compounds And Metals is, why Answer Study Guide Ionic Compounds And Metals is vital, and how to effectively learn about Answer Study Guide Ionic Compounds And Metals.
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