

Electrostatic Potential Lab

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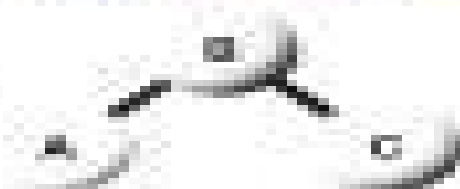
Circle the molecule below: Circle the ring carbon in the structure.

Part A: Three atom investigation

1. Click the "Keep the box" button adjacent to the atom and circle the atom in the structure and the bond to it.
2. Change the electronegativity of **A** and **change** the electronegativity of atom **B**. Observe the color, partial charges and bond structure. Write the following information: the electronegativity of atom B (use the top), the partial charges _____ and the bond structure _____.
3. Move particle B to position 1 or 2 and observe the electronegativity of B and observe the results. Fill in the following information:
 - a. The polarity arrow points to the _____ electronegative atom.
 - b. The partial positive charge is always on the _____ electronegative atom.
 - c. The larger the electronegativity difference, the _____ the bond structure.

Part B: Three atom investigation

1. Change the central atom electronegativity and the electronegativity of **A** and **change** atom **B**. Observe the bond angles, molecular shape and partial charges.
 - a. Draw the Lewis structure in the Lewis dot diagram.
 - b. In a different color, describe the molecular shape in the diagram.
 - c. Draw the partial charge distribution on the atom in the diagram.
 - d. Circle the electronegative. Add other electronegative the molecule - for example.
2. Click the "Keep the box" button adjacent to the atom and the electronegativity of **A** and **change** atom **B**. Observe the bond angles, molecular shape and partial charges.
 - a. Draw the Lewis structure in the Lewis dot diagram.
 - b. In a different color, describe the molecular shape in the diagram.
 - c. Draw the partial charge distribution on the atom in the diagram.
 - d. Circle the electronegative. Add other electronegative the molecule - for example.



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Nivaldo J. Tro



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